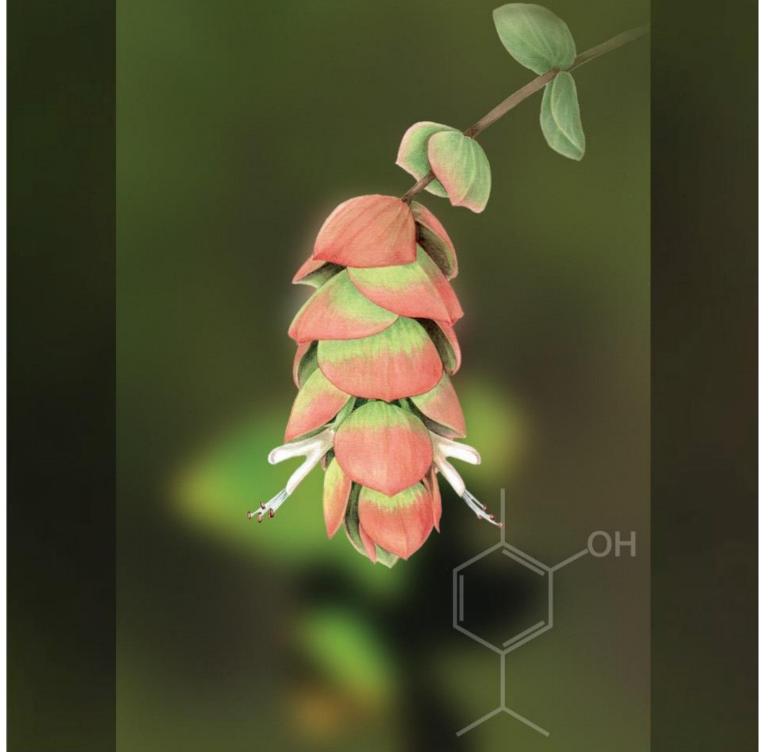
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Investigation of Streptococcus pyogenes carriage among pharmacy students in

North Cyprus

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Abstract

Streptococcus pyogenes is one of the most frequently detected bacterial agent of pharyngitis and skin infections that may result in the late complications of rheumatic fever and glomerulonephritis. The aim of the study was to investigate *S. pyogenes* carriage among pharmacy students in a university in Cyprus.

Throat samples were inoculated onto blood agar which was incubated at 37 °C for 48 hours. Gram positive, catalase negative beta hemolytic cocci which were sensitive to bacitracin and resistant to trimethoprim-sulfamethoxazole were identified as *S. pyogenes*.

A total of 140 healthy students were included in the study. 77.1% of students were Iranian, 5% each were Syrian and Iraqi, 4.3% were Nigerian and 8.6% were from other nationalities. Five (3.6%) students, all Iranian, were found to be *S. pyogenes* carriers. 4.6% of Iranian students were determined to carry *S. pyogenes*.

The study is the first study in North Cyprus reporting the low rate of group A beta hemolytic *Streptococcus* carriage in young adults in Turkish Republic of North Cyprus.

Keywords

Carriage, S. pyogenes, students, throat.

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INTRODUCTION

Streptococci which are Gram positive cocci arranged in chains or pairs are classified according to their hemolytic characteristics (alpha, beta, gamma) and to the Cpolysaccharide present in their cell wall. Although many species of Streptococcus are normal flora in various parts of human body, some serogroups of beta hemolytic streptococci, especially group А Streptococcus (GAS, S. pyogenes) and group B Streptococcus (S. agalactiae), cause important infections in human. S. pyogenes cause pyogenic infections, with a characteristic tendency to spread, as opposed to staphylococcal skin infections which are generally localized. GAS is also responsible for the non-suppurative complications, acute rheumatic fever (ARF) and acute glomerulonephritis (AGN), which may develop following upper respiratory tract and skin infections related with S. pyogenes. GAS associated diseases and complications which mainly affect children continue to have devastating effects on public health and the national prevalence GAS economy. The of pharyngitis in school aged children with sore throat was reported to be 37%. In addition to the infection it causes, GAS can also be carried in the upper respiratory tract without any symptoms. 15-32% of schoolaged children and 25% of household contacts of children with S. pyogenes

pharyngitis were reported to be GAS carriers. GAS transmits from person to person with close contact via inhalation of organisms in large droplets or by direct with respiratory contact secretions. GAS in the Carrying throat asymptomatically has been shown to have little or no effect in the development of ARF. Likewise, GAS carriers have been shown to have a minor role in transmitting the disease. However, role of GAS carriage in the development of invasive diseases has not clearly been excluded yet. Twelve of 152 household contacts of patients with invasive GAS infection were reported carry the same strain that had infected the index patient. Eradication of GAS in the carriers is not suggested by health authorities except for those who have familial history of ARF or in the presence of S. pyogenes outbreaks (Demuri and Wlad 2014; Martin 2016; Moloi 2015; Davies et al. 1996).

Cyprus has recently been an attractive country in Eastern Mediterranean to students for undergraduate and graduate studies of foreign students. Taking into account the variation in the rate of people living in different parts of the world carrying the agents of various infectious diseases, it is inevitable that such an increase in the number of foreign population in a country may lead to the change of the epidemiology of some infectious diseases locally. To our knowledge, studies related with the carriage of *S. pyogenes* in foreigners and in young adults are limited in Cyprus. The present study was undertaken to determine the rate

of GAS carriage among students from different nationalities studying pharmacy in Eastern Mediterranean University in Cyprus.

MATERIALS AND METHODS

Pharmacy students without any clinical symptoms of upper respiratory tract were included in the study. After the informed consent form had been signed, students were requested to fill out a questionnaire in order to determine the demographic characteristics of the students and the presence of any risk factors.

Throat samples were taken from the posterior pharyngeal wall and tonsils using a sterile cotton swab without swabbing the cheeks, tongues, lips or other areas of the mouth. Swabs were immediately inoculated onto Colombia agar containing 5% sheep blood (Biomérieux, France) and

media were incubated at 37 $^{\circ}$ C under the atmosphere containing 5% CO₂ for 48 hours (Spellerberg and Brandt 2015).

incubation. After dome-shaped beta hemolytic colonies, with a diameter of ≥ 0.5 mm were further identified by Gram staining, catalase test, 0.1 international unit bacitracin and trimethoprimsulfamethoxazole sensitivity. Catalase negative beta hemolytic streptococci which were found to be sensitive to bacitracin and resistant to trimethoprim-sulfamethoxazole were identified as S. pyogenes (Spellerberg and Brandt 2015).

RESULTS

In total, 140 (Confidence Level (CL) 95%, Confidence Interval (CI) 7.7%) students were included in the study. Of 140 students, 52 (47.1%) and 88 (62.9%) were male and female, respectively. Ages of students were determined to range from 17-42 and the mean age was calculated to be 21. 77.1% of students were Iranian, 5% of each were Syrian and Iraqi, 4.3% were Nigerian and 8.6% were from other nationalities. Five (3.6%) of 140 students were found to carry *S. pyogenes* in their upper respiratory tract. All of the carriers were found to be Iranian and the rate of the carriage among Iranian students was determined to be 4.6%. Demographical data related to the 5 *S. pyogenes* carriers are given in the Table 1.

Age/Sex	Living in a crowded home	Hospitalization	Use of antibiotics within 3-6 months	Previous infection with S. pyogenes	Contact with someone with <i>S.pyogenes</i> infection	Chronic disease	Immunosupression	Heavy smoker	Frequency of hand- washing in a day	Group sports	Sharing clothes and home utensils	Living in dormitory
21/Female	No	No	No	No	No	No	No	No	More than 5 times	basketball	No	No
20/Female	No	No	Yes	No	No	No	No	No	2-4times	No	No	No
20/Female	No	No	No	No	No	No	No	No	2-4times	No	No	No
21/Male	No	No	No	No	No	No	No	yes	2-4 times	No	No	No
23/Female	No	No	No	No	No	No	No	No	2-4 times	No	No	No

Table 1: Demographical data associated with five S. pyogenes carriers.

DISCUSSION

S. pyogenes is a facultative anaerobic, Gram positive coccus that is responsible for important suppurative infections and noncomplications. Pharyngitis suppurative related with S. pyogenes is still an important public health problem worldwide with an average of hundred millions of cases annually (Sanyahumbi et al. 2016). S. pyogenes carriage in children has been well investigated worldwide. In a systematic review done in 2010, the pooled prevalence of S. pyogenes carriage in throat samples of asymptomatic children younger than 18 in different countries were reported to be 12% (CL 95%; CI 9-14%) (Shaikh et al. 2010). In another systematic review including 5-15 year old children in African countries, the prevalence of carriage was reported to be 6% (CL 95%, CI 6-11%) (Moloi 2015). In a study, including a total of 1893 throat samples from 1-6 year old healthy children in 13 day-care centers, the carriage rate was reported to be 4.8% in Turkey (Sevinc and Enoz 2008). In some other studies, as high as 15–20% of the school aged children who were asymetomatic were reported to be carriers of S. pyogenes (Schwartz et al. 1981; Shulman 1994). In spite of the presence of many studies investigating the asymptomatic carrier state of GAS in children, data related with the carriage in young adults are limited. In a study performed in Poland on 205 healthy adults between 18 and 44 years old, only three (1.5%) adults were reported to carry GAS (Bura *et al.* 2016). Levy *et al.* (2015) reported the asymptomatic carriage rate of GAS as 9.6%, lower than that of our study, among students aged 18-27 years. In parallel to the result of our study, asymptomatic pharyngeal GAS colonization of adult population was reported to be less than 5% (Spellerberg and Brandt 2015; Bura *et al.* 2016).

In Cyprus, studies related with *S. pyogenes* are very limited with only one study in which the strains isolated from pharyngitis or scarlet fever cases were serotyped and their antibiotic resistance rates were reported (Koliou *et al.* 2007). To our knowledge, the present study is the first study reporting the carriage rate of *S. pyogenes* in young adults in our country.

Although the enrollment of limited number of adults is an important limitation, the present study, together with the finding of low level of carriage rate among young adults, is the first study related with GAS carriage in our region. Large scale studies are needed in the field to clarify the epidemiology of GAS carriage in Cyprus and in Eastern Mediterranean.

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Screening the cholinesterase inhibitory potential of some (1E, 4E)-1.5diphenylpenta-1.4-dien-3-one derivatives

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Abstract

 α , β -unsaturated ketones are particularly important scaffolds to be utilized in diverse organic reactions including the synthesis of many heterocyclics. Regarding their reactivities as electrophiles, their nature to be utilized as drug candidates are quite limited, although there are natural molecules with diverse pharmacological activities which are known to have α , β -unsaturated functionalization. Within this preliminary and random drug-screening based medicinal chemistry study, some (1E,4E)-1,5-diphenylpenta-1,4-dien-3-one derivatives were synthesized and their structures were identified employing chromatographic and spectral methods. The potential of the compounds to inhibit acetylcholinesterase and butyrylcholinesterase enzymes was measured employing modified Ellman's method. Although the compounds were not found to be potent inhibitors in comparison to current drugs, their activity spectra and selectivity properties displayed their availability to be utilized as important scaffolds for further design of similar α , β -unsaturated systems.

Keywords

Acetylcholinesterase, butyrylcholinesterase, α , β -unsaturated ketones.

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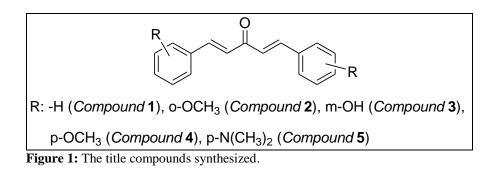
*Corresponding author: Ozan Gulcan, email: ozan.gulcan@emu.edu.tr Research Article: Volume: 2 Issue: 1 September 2019 Pages: 7-12 ©Copyright 2019 by EMUJPharmSci – Available online at dergipark.org.tr/emujpharmsci.

INTRODUCTION

Alzheimer's disease (AD) is one of the lifethreatening central nervous system diseases affecting ten millions of people worldwide (Selkoe 2001). The major symptoms of the disease include mainly the progressive cognitive decline, also referred to as dementia. Although dementia can appear throughout the scope of various disease states, AD related dementia is the major form (Whitehouse *et al.* 1982). The pathophysiology of the disease is quite complex involving diverse oxidative stress, and neurodegeneration related mechanisms (Kumar and Singh 2015).

So far, numerous scientific research studies have been conducted to discover the exact pathophysiology and epidemiology of AD. However, no single biochemical pathway has solely been attributed to be involved within the generation of the disease. It is known that cholinesterase system, particularly the action of acetylcholine on some muscarinic and nicotinic receptors, is an important tool of cognition, involving personal characteristics, recognition, learning, and daily activities (Ferreira-Vieira et al. 2016). Using this information, cholinesterase inhibition mechanism has been suggested to the clinic through the end of the last century to overcome with dementia related symptoms of AD. Indeed, acetylcholine hydrolyzing enzymes, acetylcholinesterase (AChE) and

butyrylcholinesterase (BuChE), have been targeted and clinically used current drugs obtained. Donepezil, have been rivastigmine, and galantamine are the currently used cholinesterase inhibitor drugs used for the treatment of cognition symptoms of AD (Deardorff WJ et al. 2015). Based on the pharmacokinetic and pharmacodynamic variances among these drugs, there has been a continuous interest on the discovery of novel cholinesterase inhibitor molecules. Regarding this point, within this preliminary random screening base study, we have synthesized some (1E, 4E)-1,5-diphenylpenta-1,4-dien-3-one derivatives and screened their potential to inhibit AChE and BuChE. Within the scope of Organic Chemistry II lectures in Eastern Mediterranean University, pharmacy students exploit from Aldol reaction to learn and practice the synthesis of (1E,4E)-1,5diphenylpenta-1,4-dien-3-one. Since the methodology employs the reaction between benzaldehyde and acetone, we have used several substituted benzaldehyde derivatives. The title compounds are shown in Figure 1.



MATERIALS AND METHODS

Benzaldehyde, o-anisaldehyde, panisaldehyde, 3-hydroxybenzaldehyde, 4dimethylaminobenzaldehyde, acetone, sodium hydroxide and ethanol were obtained from Sigma Aldrich (CA, USA). Purities of the chemicals were more than 99% as stated on their labels. Therefore, no other purification was conducted on the reagents.

Synthesis of the title compounds

For a typical reaction, 62.5 mmol of sodium hydroxide was dissolved in water-ethanol solution (55:45) in a 100 ml reaction flask. 24.5 mmol of the benzaldehyde derivative and 12.9 mmol of acetone was added to the solution. The reaction was stirred at room temperature for 15 min. The precipitate formed was filtered off and washed with acidified aqueous.

Structure identification and characterization

The reactions were monitored employing Thin Layer Chromatography (TLC) (Alugram Xtra SIL G/UV₂₅₄ 0,2 mm silica gel 60 with fluorescent indicator from Germany) with an n-hexane ; ethyl acetate (1:1) mobile phase. The infrared spectra of the compounds were obtained with a Shimadzu FT-IR Prestige Infrared Spectrophotometer. The ¹H-NMR and ¹³CNMR spectra of the compounds were obtained using a Bruker 400 NMR Spectrophotometer. Trimethylsilane was used as internal standard and DMS-d6 as solvent.

(1E,4E)-1,5-diphenylpenta-1,4-dien-3-one

(compound 1): Yield 85%, solid yellow crystals, mp 125°C (uncorrected data). IR, 3055 (ArC-H), 1648 (-C=O). ¹HNMR, 7.87 ppm (d, HC=<u>CH</u>-Ar), 7.81 ppm (dd, Ar-H, o), 7.48 ppm (dd, Ar-H, m), 7.42 ppm (s, Ar-H, p), 7.38 ppm (d, -<u>CH</u>=CH-Ar). ¹³CNMR, 188ppm (C=O).

(1E,4E)-1,5-bis(2-methoxyphenyl)penta-

1,4-dien-3-one (compound 2): Yield 87%, solid pale yellow crystals, mp 134°C (uncorrected data). IR, 3027 (ArC-H), 1668 (-C=O). ¹HNMR, 7.93 ppm (d, -CH=<u>CH</u>-Ar), 7.81 ppm (d, Ar-H, o), 7.42 ppm (t, Ar-H, p), 7.28 ppm (d, Ar-H, m), 7.09ppm (d, Ar-H, o'), 7.09ppm (d, -<u>CH</u>=CH-Ar). ¹³CNMR, 188ppm (C=O).

(1E,4E)-1,5-bis(3-hydroxyphenyl)penta-

1,4-dien-3-one (**compound 3**): Yield 89%, solid brown crystals, mp 136°C (uncorrected data). 3075 (ArC-H), 1620 (-C=O). ¹HNMR, 9.74 ppm (s, Ar-O<u>H</u>), 7.79 ppm (d, -CH=<u>CH</u>-Ar), 7.35 ppm (t, Ar-H, m), 7.31 ppm (d, Ar-H, o), 7.28 ppm (d, -<u>CH</u>=CH-Ar), 7.25 ppm (d, Ar-H, p), 6.95 ppm (d, Ar-H, o). ¹³CNMR, 188ppm (C=O).

(1E,4E)-1,5-bis(4-methoxyphenyl)penta-

1,4-dien-3-one (**compound 4**): Yield 81%, solid yellow crystals, mp 128°C (uncorrected data). 3033 (ArC-H), 1648 (-C=O). ¹HNMR, 7.73 ppm (d, -CH=<u>CH</u>-Ar), 7.68 ppm (d, Ar-H, o), 7.16 ppm (d, -<u>CH</u>=CH-Ar), 6.99 ppm (d, Ar-H, m), 3.79 ppm (s, Ar-O<u>CH₃</u>). ¹³CNMR, 188ppm (C=O).

(1E,4E)-1,5-bis(4-(dimethylamino)

phenyl)penta-1,4-dien-3-one (**compound 5**): Yield 85%, solid orange crystals, mp 141°C (uncorrected data). 3038 (ArC-H), 1634 (-C=O).). ¹HNMR, 9.69 ppm (s, N-H), 7.69 ppm (d, -CH=<u>CH</u>-Ar), 7.60 ppm (d, Ar-H, o), 7.05 ppm (d, -<u>CH</u>=CH-Ar), 6.74 ppm (d, Ar-H, m), 3.48 ppm (s, N-<u>CH</u>₃). ¹³CNMR, 188ppm (C=O).

Determination of AChE and BChE inhibitory activities

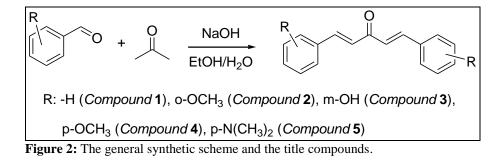
The modified spectrophotometric method of Ellman was used to determine AChE and BuChE inhibitory activities of 5 compounds synthesized (Gulcan *et al.* 2014). The enzymes used for cholinesterase activity studies were, electric eel AChE (eeAChE) (Sigma) and equine BuChE (Sigma).

Acetylthiocholine iodide and butyrylthiocholine chloride (Sigma, St. Louis, MO, USA) were employed as substrates of the reaction. 5, 5'-Dithio-bis (2nitrobenzoic) acid (DTNB, Sigma, St. Louis, MO, USA) was used for the measurement of the cholinesterase activity. 50 mM Tris HCl buffer (pH 8.0), 6.8 mM DTNB, 2 µl of sample solutions and 10 µl of AChE/BChE solution were added in a 96-well microplate. The reaction was then initiated with the addition of 10 µl of acetylthiocholine iodide/butyrylthiocholine chloride. The hydrolysis of acetylthiocholine iodide/butyrylthiocholine chloride was monitored by the formation of the yellow 5thio-2-nitrobenzoate anion as a result of the reaction of DTNB with thiocholines, catalyzed by enzymes at a wavelength of 412 nm utilizing a 96-well microplate reader (Varioskan Flash, Thermo Scientific, USA) and incubated for 15 min at 27°C. The and calculations measurements were evaluated by using SkanIt Software 2.4.5 RE for Varioskan Flash software. Percentage of inhibition of AChE and BuChE were determined by comparison of the rates of reaction of samples relative to blank sample (methanol) using the formula $(E-S)/E \ge 100$, where E is the activity of enzyme without the test sample and S is the activity of enzyme with the test sample. The experiments were

done in triplicate. Donepezil hydrochloride and rivastigmine were used as reference compound. The percent inhibition at 100 and $50 \,\mu M$ was obtained for each test compound with standard compounds.

RESULTS AND DISCUSSION

The general synthetic scheme of the title compounds is shown in the Figure 2. Employing the general synthetic scheme (1E, 4E)-1,5-diphenylpenta-1,4-dien-3-one derivatives were synthesized. Following the structure identification studies, the compounds were screened for their potential to inhibit AChE and BuChE enzymes. The results obtained are displayed the Table 1.



	% Inhibition	% Inhibition	% Inhibition	% Inhibition
Compounds	(AChE)	(AChE)	(BuChE)	(BuChE)
	50µM	100µM	50µM	100µM
Compund 1	36.54 ± 0.33	54.2 ± 0.23	45.90 ± 0.29	58.6 ± 0.41
Compund 2	11.1 ± 0.04	25.6 ± 0.09	21.52 ± 0.04	36.85 ± 0.04
Compund 3	29.77 ± 0.36	45.04 ± 0.37	49.65 ± 0.31	68.11 ± 0.72
Compund 4	29.01 ± 0.12	44.37 ± 0.28	25.76 ± 0.13	39.37 ± 0.91
Compund 5	14.51 ± 0.07	24.75 ± 0.14	18.31 ± 0.08	39.66 ± 0.21
Rivastigmine	65.02 ± 0.11	76.12 ± 0.06	78.31 ± 0.03	83.61 ± 0.09
Donepezil	91.20 ± 0.17	94.19 ± 0.10	83.75 ± 0.04	88.22 ± 0.62

Table 1: The potential the title compounds to inhibit AChE and BuChE enzymes.

According to the results, the title compounds were not found to be superior to the currently used drugs donepezil and rivastigmine, which were already used as references in this study. However, each title compound displayed activity for both AChE and BuChE. Although it is not apparent for each title molecule, a tendency for more inhibition of BuChE particularly for compounds 2, 3, and 5, was identified, it is noteworthy to state that it is very critical to identify IC_{508} of the title compounds to

further proof this observation. Besides, the results were primitive to describe a structure activity relationship study. In other words, the (1E, 4E)-1,5-diphenylpenta-1,4-dien-3one main moiety was obtained important for activity, but no net result was observed depending on the substitutions followed. From this point of view, the results of the study indicated that this scaffold stands a good candidate to design novel (1E,4E)-1,5diphenylpenta-1,4-dien-3-one derivative potent cholinesterase inhibitors. However, more data related to the number of more substitutions and concomitant IC₅₀ are needed to explore both the moleculereceptor interactions and a concessive structure activity relationship studies.

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Outer and inner morphological characteristics of Saturea and Thymbra taxa

exported as Oregano from Turkey II

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Abstract

Oregano, a collective name of a group of taxa that has been used as spice. Different genera and species of the family Lamiaceae that contain thymol and carvacrol are being used as oregano. In Turkey, there are 9 taxa that belong to 3 genera which are being exported as oregano. In the outer and inner morphological characteristics of the taxa *Satureja thymbra* L., *Satureja cuneifolia* Ten., *Thymbra capitata* (L.) Cav. and *Thymbra spicata* L. subsp. *spicata* which are exported as oregano have been reported. Distribution maps of the four taxa have been presented based on the examined specimens kept in the 15 herbaria. The specimens are listed in the appendix and local usage and vernacular names are shown as tables.

Keywords

Anatomy, morphology, oregano, Satureja, Thymbra, Turkey.

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Article Info

Oregano is a collective name of a group of taxa in which different genera and species of the family Lamiaceae that contain thymol and carvacrol are present. These taxa are used as spice and known as "kekik" in Turkish and are very important exported plants in Turkey. Oregano has been used in Anatolia since the 7th century BC as a spice and for medical treatment (Sadikoglu and Ozhatay 2015). Various studies observed the position of these plants in nature and the place of oregano in the trade (Kirimer et al. 2003; Duzenli and Karaomeroglu 2003; Tumen et al. 2003; Gemici 2003). Oregano species are collected in Turkey and their trade is an important source of income.

In Turkey, 9 taxa are being exported as oregano. These are: Origanum minutiflorum O. Schwarz & P.H. Davis, O. majorana L., O. onites L., O. syriacum L. subsp. bevanii (Holmes) Greuter & Burdet, O. vulgare L. subsp. hirtum (Link) Ietsw., Satureja thymbra L., S. cuneifolia Ten., *capitata* (L.) Cav. Thymbra (Syn. Coridothymus capitatus (L.) Rchb.f.) and T. spicata L. subsp. spicata. The taxa that belong to the genus Origanum are roundleafed where as those that belong to Satureja and Thymbra are acute-leafed. The major genus Origanum taxa have recently been published (Sadikoglu and Ozhatay 2015). In this paper, outer and inner morphological characteristics of exported

Satureja and Thymbra taxa have been presented together with their local name, distribution usage and in Turkey. *Coridothymus* was represented by only one taxon in Turkey. It was a monotypic genus but according to the latest systematic changes Coridothymus capitatus (L.) Rchb.f. is called as *Thymbra capitata* (L.) Cav. (Güner et al. 2012). Flavonoid analysis supports the close relationships of a taxon, such as the *Coridotyhmus capitatus* taxa, also known as *Thymus capitatus* (L.) Hoffmgg. & Link and Thymbra capitata (L.) Griseb. According to micromorphological and caryological data, it has been suggested to be named as Thymbra The presence of flavonoid capitate. aglycones in this taxon is the same as those of other Thymbra species, but different from those of the *Thymus* species (Barberan and Gil 1992). According to another study difference that investigates between Thymus and Thymbra species, it was determined that the naming as Thymbra capitata was not accurate because of the convexity of the calyx and the absence of the 13-veins in the calyx (Tanker and Ilisulu 1981). In Turkey, the genus Satureja is represented by 15 taxa and 5 of them are endemic. The genus *Thymbra* is represented by 5 taxa and 1 of them is endemic (Guner et al. 2012). Endemic or rare taxa are also sold in the open markets for various usages.

MATERIALS AND METHODS

Flowering and fruiting specimens were collected in Adana, Antalya, Balıkesir, Burdur, Canakkale, Hatay, Isparta, Izmir, Kirklareli, Malatya, Mersin, Mugla, Van Yalova provinces during Juneand September periods in 2001-2004. Specimens were identified by Narin Sadikoglu and kept in the Herbarium of Istanbul University Faculty of Pharmacy (ISTE).

ADA, AEF, ANK, EGE, ESSE, GAZI, HUB, ISTE, ISTF, IZEF, MARA, MARE, MUFE, VANF and INU herbaria were visited. The taxa that are exported and used as oregano in Turkey were investigated in detail. All examined specimens were the presented in appendix. Oregano specimens collected and fixed in 70% ethanol since 1996 were the material of the anatomical part which are housed in ISTE. Morphological drawings were made by **SZH10** stereomicroscope. Olympus Anatomical drawings were made by Leitz SM-LUX binocular microscope with Leitz Weitzler drawing tube. Photographs were taken by Olympus BH2 trinocular

Anatomical sections photomicroscope. were taken by hand from the leaves of the plants. Leaves are cleared by using 15% sodium hypochlorite for 15 minutes. Sections were stained in 1% safranin and mounted in chloral hydrate: glycerol: water (8:2:1). For transverse sections, leaves were placed in Jeffrey's solution (10% aqueous nitric acid and 10% aqueous chromic acid [1:1]) for 1 week at room temperature until the epidermis began to separate (Berlyn and Miksch 1976). Segments of epidermis with the attached outermost cell layer of cortical parenchyma were collected, washed in water and the tissue is mounted.

Stomata amount per mm² were established and stomata index both upper and lower surfaces were calculated for using the formula of SI = (Stomata amount / Epiderm cell + Stomata amount) x 100. Averages of the counts were determined for 10 specimens obtained from 10 plants for 10 views of each. Palisade ratio and areoles per mm² were also established for the preparations of powdered oregano (Sener *et al.* 1985).

RESULTS

According to our observations, the exported *Satureja* and *Thymbra* taxa were distributed in the west, south and north of Turkey and they are used as spice, herbal tea and remedies. Local usages of each taxa are listed in the Table 1.

Scientific name	Local name	Local usage
Satureja thymbra	Sahil sivrisi, kılıç kekiği, sivri kekik, taş kekiği	Spice
S.cuneifolia	Sivri kekik, kılıç kekik, yayla kekiği, dağ kekiği, aş kekiği, taş kekiği, kaya kekiği, çorba kekiği, yabani kekik	Oil, spice, herbal tea
Thymbra capitata	Timari, sivri kekik	Spice, herbal tea
T. spicata subsp. spicata	Zahter, sivri kekik, at kekiği, mor kekik, aş kekiği	Skin disorders, sedative, analgesic, cold, antitussive, antirheumatismal, hyperglisemia, antihypertansive, hypercholesterolemia, gastrointestinal disorders, fuel, spice, herbal tea

Table 1: Local names and usage of *Satureja* and *Thymbra* taxa that are exported as oregano.

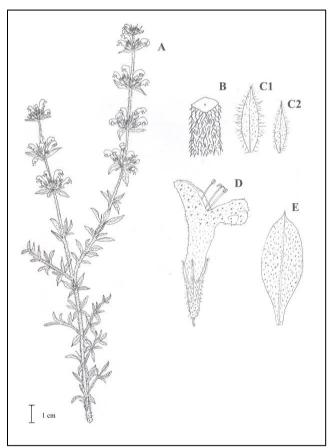
Outer morphological characteristics and distribution

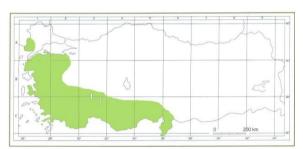
1. Satureja thymbra L.

Type: Described from Crete (Hb. Linn. 723/2 photo!).

Main diagnostic morphological characteristics are:

Bracteoles ovate, acuminate-aristate, long-ciliate; leaves linear- to ovate-spathulate; calyx actinomorphic; corolla mauve or purple, 8-12 mm (Figure 1). The distrubution of *Satureja thymbra* is shown in Map 1.





Map 1: Distribution of *Satureja thymbra* in Turkey.

Figure 1: *Satureja thymbra* (ISTE 19137). A. general view, B. stem, C. bract, D. flower, E. leaf.

Flowering period: April-July

Habitat: Dry scrub, especially calcareous phrygana, sl.-400 m.

Distribution in Turkey: West and South Turkey (Map 1)

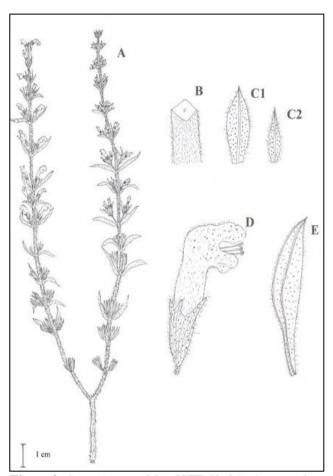
General distribution: Sardinia, Greece, Aegean, W. Syria, Cyprus, Cyrenaica. East Mediterranean element.

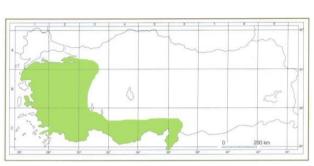
2. Saturea cuneifolia Ten.

Type: Described from Italy (holo. NAP).

Main diagnostic morphological characteristics are:

Bracteoles linear, not ciliate; leaves cuneate towards base, broadest towards apex; calyx subbilabiate nearly to middle; corolla mostly white, 6-8 mm (Figure 2). The distrubution of *Satureja cuneifolia* is shown in Map 2.





Map 2: Distribution of *Satureja cuneifolia* in Turkey.

Figure 2: *Satureja cuneifolia* (ISTE 52634). A. general view, B. stem, C. bract, D. flower, E. leaf.

Flowering period: July-August

Habitat: Rocky slopes, on schist and limestone, cliffs, 300-2000 m.

Distribution in Turkey: West and South Turkey (Map 2)

General distribution: Spain, Italy, Jugoslavia, Albania, Greece, Lebanon, N. Iraq, Turkey. Mediterranean element.

3. Thymbra capitata (L.) Cav. Syn. Coridothymus capitatus Rchb.f.

Type: Described from Baetica (Andalucia), Crete, Hispalis (Seville) and Greece (Hb. Linn. 723/11 photo!).

Main diagnostic morphological characteristics are:

Calyx tube dorsally compressed, with two ciliolate flanges. Inflorescence capitate; calyx 20-22-veined; leaves 4-10 mm, subtriquetrous (Figure 3). The distrubution of *Thymbra capitata* is shown in Map 3.

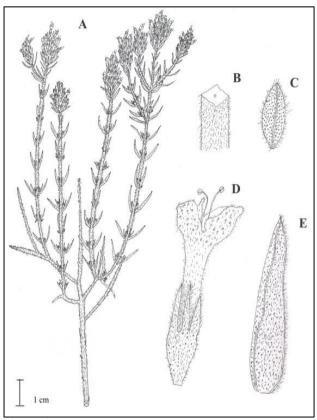


Figure 3: *Thymbra capitata* (L.) Cav. (ISTE 81671). A. general view, B. stem, C. bract, D. flower, E. leaf.

Map 3: Distribution of *Thymbra capitata* in Turkey.

Flowering period: May-October

Habitat: Costal phrygana, open macchie, s.l.-1400 m.

Distribution in Turkey: West Turkey, South Anatolia (Map 3)

General distribution: Greece, Italy, Turkey. Mediterranean element.

4. Thymbra spicata L. subsp. spicata

Type: Described from Macedonia: Libano (Hb. Linn. 724/1 photo!).

Main diagnostic morphological characteristics are:

Calyx tube dorsally compressed, with two ciliolate flanges. Inflorescence spicate; calyx 13-veined; leaves to 20 mm, conduplicate (Figure 4). The distrubution of *Thymbra spicata* is shown in Map 4.

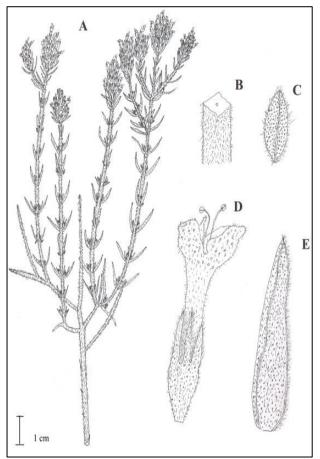


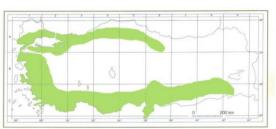
Figure 4: *Thymbra capitata* (L.) Cav. (ISTE 81671). A. general view, B. stem, C. bract, D. flower, E. leaf.

Flowering period: June-July

Habitat: Dry often rocky places (usually calcareous), in scrub, phrygana, steppe, s.l.-1000 m.

Distribution in Turkey: North, West and South Turkey, SE Anatolia (Map 4)

General distribution: Greece, Aegean, Cyprus, W. Syria, N. Iraq. East Mediterranean element.



Map 4: Distribution of *Thymbra spicata* subsp. *spicata* in Turkey.

1. Satureja thymbra L.

On the superficial section, both upper and lower epidermal cells are straight. Hairs have cuticular patterns, 1-2 celled, contain crystals very densely. Labiatae type glandular hairs per cm² are 400-(630)-1200 on epidermis. Stomata occur on both upper and lower epidermis (amphistomatic) that are usually in same density. Stomata amount is established per mm² as 800-(1090)-1500 on the lower surface and as 500-(850)-1500 on the upper surface. On the cross-section, guard cells are on the same level with the epidermis cells (mesomorphic stomata). Subsidiary cells are 2. Stomata index is established 95.11 for the lower surface and 94.33 for the upper surface. Palisade ratio in powdered specimens is established 4.5, areoles per mm² is 7. Under the upper epidermis 4-5 layered collenchyma tissue is present (Figure 5).

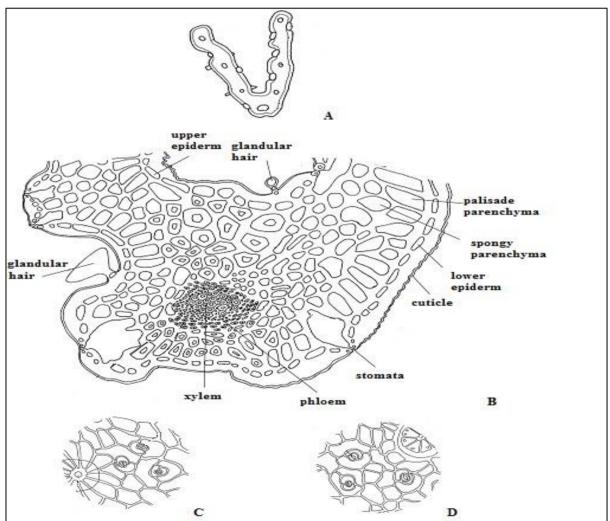


Figure 5: *Satureja thymbra* (Narin/kekik/051). A. Cross-section of leaf (x80), B. Middle vessel (x450), C. Lower surface of epiderm (x225), D. Upper surface of epiderm (x225).

2. Satureja cuneifolia Ten.

On the superficial section upper epidermal cells are straight, lower ones are slightly undulate. Hairs are 1-4 celled, contain crystals very densely. Labiatae type glandular hairs per cm² are 400-(710)-1000 on epidermis. Stomata occur on both upper and lower epidermis (amphistomatic) and are denser on the upper surface. Stomata amount is established per mm²: 600-(1030)-

1700 on the lower surface and, 1300-(2120)-3200 on the upper one. On the crosssection, guard cells are on the same level with the epidermis cells (mesomorphic stomata). Subsidiary cells are 2, rarely 3, very rarely 4. Stomata index is established 92.29 for the lower surface and as 96.14 for the uppers. Palisade ratio in powdered specimens is established as 3.35 and areoles per mm² is 5.17 (Figure 6).

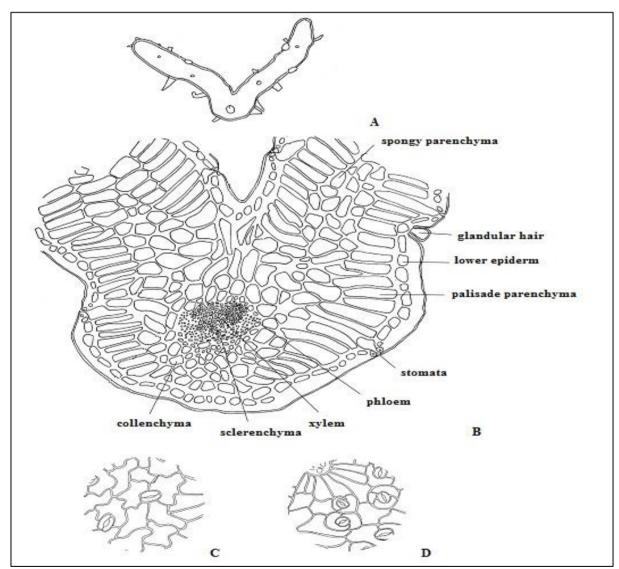


Figure 6: *Satureja cuneifolia* (Narin/kekik/051). A. Cross-section of leaf (x80), B. Middle vessel (x450), C. Lower surface of epiderm (x225), D. Upper surface of epiderm (x225).

3. Thymbra capitata (L.) Cav.

On the superficial section both upper and lower epidermal cells have straight walls. Epidermis cells of the upper surface are slightly bigger than those of the lower surface. Hairs have cuticular patterns, are 1-3 celled, and contain crystals very densely. Labiatae type glandular hairs per cm² are 400-(590)-1200 on epidermis. Stomata occur on both upper and lower epidermis (amphistomatic) and are denser on the surface. Stomata upper amount is established per mm^2 as 300-(1070)-1600 on the lower surface and as 600-(1230)-1700 on the upper surface. On the cross-section,

guard cells are on the same level with the epidermis cells (mesomorphic stomata). Subsidiary cells are 2. Stomata index is established as 92.48 for the lower surface 92.76 for the uppers. Palisade and parenchyma is 1-layered and spongy parenchyma is 1-3 layered. Palisade ratio in powdered specimens is established as 3.8, areoles per mm^2 is 4.25. Below the lower and upper epidermis, 1-layered palisade parenchyma and 1-3 layered irregular parenchyma cells are present. A few collenchyma cells were observed on the lower surface (Figure 7).

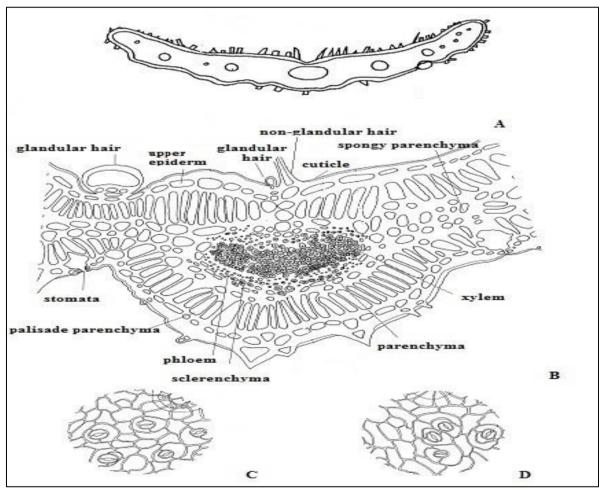


Figure 7: *Thymbra capitata* (Narin/kekik/097). A. Cross-section of leaf (x60), B. Middle vessel (x450), C. Lower surface of epiderm (x225), D. Upper surface of epiderm (x225).

4. Thymbra spicata L. subsp. spicata

On the superficial section, the walls of the upper epidermal cells are thicker than those of lower epidermal cells and both of them are straight. Epidermis cells of the upper surface are bigger than the lowers. Hairs have cuticular patterns, are 1-3 celled, and contain crystals very densely. Labiatae type glandular hairs per cm² are 1600-(2150)-2700 on epidermis. Stomata occur on both upper and lower epidermis (amphistomatic) that are denser on the upper surface. Stomata amount is established per mm² as 700-(1190)-1700 on the lower surface and, 400-(1180)-2100 on the upper one. On the cross-section, guard cells are upper than

the epidermis cells (higromorphic stomata). Subsidiary cells are 2. Stomata space is evident. Stomata index is established as 95.35 for the lower surface and as 94.93 for the uppers. Palisade parenchyma is 1layered, spongy parenchyma is 1-2 layered. Palisade ratio in powdered specimens is established as 4.8, areoles per mm² is 3.85. Below the lower and upper epidermis, 1layered palisade parenchyma and 1-3 layered irregular parenchyma cells are present. Less sclerenchymatous tissue occur on the upper and lower of midrib. Tracheary elements and thin walled parenchyma cells between them are on the xylem (Figure 8).

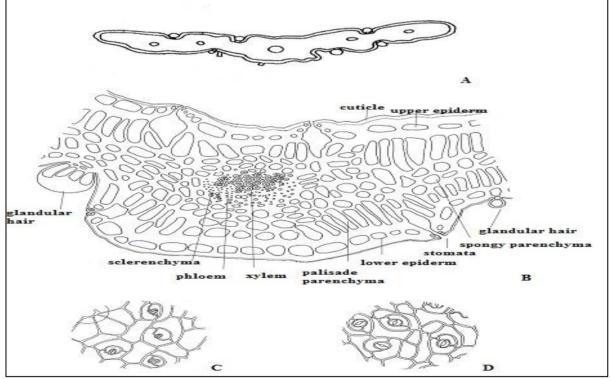


Figure 8: *Thymbra spicata* L. subsp. *spicata* (Narin/kekik/053). A. Cross-section of leaf (x75), B. Middle vessel (x450), C. Lower surface of epiderm (x225), D. Upper surface of epiderm (x225).

DISCUSSION

The findings of studied taxa are in compatible with the literature (Kaya et al. 1994; Tanker and Ilisulu 1981; Erken 2001). The Thymbra spicata subsp. spicata taxa collected from many populations were shown to reveal high morphological polymorphism. It is determined that these are not local. The differences of these taxa would be clarified by biosystematical investigations. Leaves are isobilateral. In all species lateral veins are not prominent. Anatomically, the lateral veins have the same structure with the midrib but the vascular bundles are reduced. S. thymbra has thick walls, 1-3 celled hairs and much prominent cuticular S. patterns. In cuneifolia, hairs are 1-3 celled and has thick walls, there are no cuticular patterns, and subsidiary cells are 2-3 or 4. In T. capitate, hairs are 1-3 celled and prickle hairs are dense and the outlines of the anticlinal walls are straight. Differences in morphological characteristics of examined *Satureja*, and *Thymbra* taxa are summarised in the Table 2. *T. spicata* subsp. *spicata* has sparsely hairs and straight anticlinal walls, subsidiary cells are smaller on the lower surface. *T. spicata* subsp. *spicata* is the species which has the densest glandular hairs and the sparsest non-glandular hairs.

T. capitata, has sparse glandular hairs. The glandular hairs are smallest in both *S. thymbra* and *T. capitata* species. *S. thymbra* has the densest cuticular patterns. *S. cuneifolia*, has the densest stomata amount on the upper surface. The crystals are densest in *T. spicata* subsp. *spicata* and *T. capitata* species. The differences of anatomical characteristics are summarised in the Table 3.

	Satureja thymbra	Satureja cuneifolia	Thymbra capitata	Thymbra spicata subsp. spicata	
Colour (dried and powdered)	AND AND AND AND AND AND AND AND AND AND	and the free way		Subsp. Spicau	
	Brownish green	Dark green	Light green	Green-purple	
Hairs	Minutely recurved	Minutely recurved	Canescent	Minutely recurved on two opposite sides	
Leaves	9-9.5×2.8-5.5 mm; cuneate-oblanceolate; apex acute, mucronate; entire, conduplicate; lateral veins not prominent	3-15×1-8 mm; cuneate- oblanceolate; apex acute, mucronate, subobtuse; entire, conduplicate; lateral veins not prominent	6-10×1-1.2 mm; triangular-linear; apex acute, entire, deltoid, lateral veins not prominent	3-15×1-3 mm; linear, linear-lanceolate; apex acute, entire, conduplicate; lateral veins not prominent	
Inflorescence	Distant verticillate	Spicate	Capitate	Spicate	
Bracts	Oblong-elliptic	Linear-lanceolate	Ovate	Elliptic-lanceolate	
Calyx	Tubular-campanulate, 2- lipped	Tubular-turbinate, slightly 2-lipped	Dorsaly flattened, 2-lipped	Dorsaly compressed, 2- lipped	

Table 2: Morphological differences of examined Satureja and Thyn	<i>mbra</i> taxa.
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Table 3: Anatomical differences of leaves of Satureja and Thymbra taxa.

Characteristics	Satureja thymbra	S.cuneifolia	Thymbra capitata	T.spicata subsp. spicata
Palisade ratio	4.5	3.35	3.8	4.8
Areoles per mm ²	7	5.17	4.25	3.85
Stomata amount per mm ² (L/U)*	1090/850	1030/2120	1070/1230	1190/1180
Stomata index (L/U)*	95.11/94.33	92.29/96.14	92.48/92.76	95.35/94.93
Glandular hairs (no stalk) per mm ²	630	710	590	2150
Subsidiary cells	2	2-4	2	2
Collenchyma (lower epidermis	2	2	Several cells one by one	-
Collenchyma (upper epidermis)	4-5	4-5	Several cells one by one	-
Palisade parenchyma	2-layered	2	1	1
Spongy parenchyma	1-2	1-2	1-2	1-2
Hairs	1-2	1-4	1-3	1-3
Glandular hairs	Head 1, stalk 1-celled	Head 1, stalk 1-celled	Head 1, stalk 1-celled	Head 1, stalk 1-celled
Sclerenchyma	1	2-3	1	1
Collenchyma (corner)	8	5-7	5-6	6-7
Collenchyma	1-3	1-3	2-3	2-3
Floem	3-5	8	5-6	3-4
Cuticle (L/U)*	Smooth, slightly undulate / smooth	Slightly undulate / smooth	Smooth / smooth	Smooth / smooth
Epidermis cells (L/U)*	Big / small	Big / small	Small / big	Small / big

* (L/U) : Lower/upper

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Appendix

Examined specimens;

Satureja thymbra L.

A1(E) Edirne: Kaya kö., Kınak mevkii, 26.v.1996, G. Tümen (ESSE 12805)! B1 Balıkesir: Edremit-Avcılar kö, 250 m, 9.v.1966, (EGE 5291)! Alibey adası, Alibey Tepe batısı, 180 m, 26.v.1997, K. Alpınar (ISTE 74190)! İzmir: Makilik, Sığacık, Seferihisar, 15.v.1993, E. Saver (IZEF 1150)! d. Seferihisar, Sığacık-Teos, 15.v.1993, N. Zeybek (IZEF 1233)! Bergama, 26.vi.1977, İ. Akbulut (IZEF 2172)! Kemalpaşa, Kızılüzüm kö., 200 m, 27.v.1996, E. Saver (IZEF 4061)! Seferihisar-Terkos, ca. 0-50 m, 16.v.1986, (EGE 21800)! Cimentas, 15.iv.1966, (EGE 5085)! Hills N of Bornova, 2.v.1969, (EGE 7884)! Hills NE of Bornova, 15.v.1969, (EGE 4305)! Cesme, 7. iv. 1968, (EGE 3336)! Balikliova-Mordoğan, 5. i. 1966, (EGE 5431)! Manisa: Ücpinar-Gediz, Muhtarlık özel alanı, 06.v.1995, U. Zeybek (IZEF 3296)! Soma, orta istasyondan 200 m yukarısı, ca. 400 m, 12.v.1977, (EGE 26032)! C1 Aydın: Ortaklar-Çamlık, 300 m, 12.v.1967, (EGE 6359)! Kuşadası yolu üzeri, 6.xı.1967, (EGE 4690)! Kuşadası, Kalamaki deresi, c. 200 m, 12.v.1968, (EGE 6109)! Paşa yaylası, c. 250 m, 27.ıx.1965, (EGE 6051)! Samsundağı, above Güzelçamlı, 29.v.1969, (EGE 4407)! Akköy, Söke-Didim, Yenihisar'a 5 km kala, 3.Iv.1995, (EGE 18996)! Germencik-Ortaklar kasabası, Öğretmen Lisesi çevresi, 27.v.1993, G. Tümen (ESSE 10479, 10487)! Kuşadası-Davutlar Milli parkı, kanyan bölgesi, 13.v.1995, N. Öztürk (ESSE 11701)! Kuşadası yolu, Kuşadası yakını, kayalık sırtlar, 9.1v.1971, A. ve T. Baytop (ISTE 19137)! Didim yolu, Didim'e 10 km kala, maki arası, 10.1v.1971, A. ve T. Baytop (ISTE 19192)! Dilek Yarımadası, 100 m, 11.v1.1982, M. Miski, E. Bütün (ISTE 48944)! Izmir: d. Selçuk, 3-4 km N Yoncaköy, 250 m, 26.v.1993, U. Zeybek (IZEF 1219)! Samsun D., Güzel çamlı üstleri, 520 m, 5.vıı. 1989, E. Tuzlacı (MARE 1970)! Muğla: Bodrum-Mumcular, 01.1.1950, M. Polat (IZEF 1757)! Datça, 19.vii.1966, (EGE 6034)! Bodrum, Gölköy yakını, Cennet koyu çevresi, E. Tuzlacı (MARE 8540)! Yalıkavak-Bodrum yolu 3. km, yol kenarı, 10.v.1995, N. Öztürk (ESSE 11699)! Datça, Bozdağ (Kocadağ), Mesudiye kö. üstleri, 600 m, 3.v11.1983, E. Tuzlacı (ISTE 51504)! Datça- Marmaris, Datça'dan 13 km, Gebekum mevkii, d.s. kumullarda, 5.vn.1983, E. Tuzlacı (ISTE 51589)! Datça, Kocadağ kuzey etekleri, Karaköy yakını, d.s., 12.v.1984, E. Tuzlacı (ISTE 53318)! Datça-Knidos, Datça'dan 10 km, kireçtaşlı sırtlar ve tepelerin rutubetli kuzey yüzü, 5.1v.1995, A.J. Byfield, D. Pearman B 1501 (ISTE 68850)! C2 Antalya: Kaş 5 km, 26.v11.1960, Khan et all. 188 (ANK)! Burdur: Dirmil-Fethiye, 51 km S Dirmil, 1000 m, 20.v1.1981, Max Nydegger 16330 (HUB 22691)! Kas, 22.v.1967, T. Baytop (ISTE 12280)! Kas Fethiye yolu, Fethiye'ye 90 km, 22.iv.1978, A. ve T. Baytop (ISTE 39063)! Kalkan Kas yolu, Kalkan yakını, Kalkan'a 3 km, 120 m, 5.v.1980, A. ve T. Baytop, A. Attila, N. Sütlüpinar (ISTE 44163)! Muğla: d. Marmaris, 6 km W Resadive, 15.v.1997, E. Saver, N. Zeybek (IZEF 4929)! Fethiye, Kemer bucağı, Derecatı kö., Bayaslar mevkii, c. 300 m, 15.vi.1967, (EGE 5812)! Fethiye, Bayaslar-Derecati, 15.vi.1967, (EGE 4913)! Fethive, Hisarönü kö., Belceğiz mevkii, 250 m, 11.v.1967, (EGE 4959, 4935)! Fethiye, Kemer-Taşocağı, 10.vii.1966, (EGE 6073, 6061, 6004)! Köyceğiz, Ekincik-Marina, 20 m, kızılçam ormanı, metamorfik kalkerli arazi, 17.1v.1991, A. Güner 8785, M. Vural, H. Duman, AA. Dönmez, B. Mutlu (HUB 2293)! Köyceğiz, Yangı kö., Yangı de., 100-200 m, sarp ve derin vadi, kalkerli kayalıklar, 18.IV.1991, A. Güner 8888, M. Vural, H. Duman, AA. Dönmez, B. Mutlu (HUB 2292)! Marmaris, M. Milli Parki, 150 m, serpentin, 29.v1.1997, H. Şağban 1868 (HUB 22381)! Marmaris-Datça yolu, yamaçlar, 16.v.1987, K.H.C. Başer (ESSE 8183)! Gökova inişi, yol kenarı, yamaçlar, 15.v.1987, K.H.C. Başer (ESSE 7809)! Fethiye, Akbel, Çöğmen kö., 15.x1.1992, G. Tümen (ESSE 9984)! Fethiye, 10.v1.1990, G. Tümen (ESSE 8949)! Marmaris-Datça, Çubucak orman kampı çevresi, viii.1992, G. Tümen (ESSE 9820)! Fethiye, Söğütlüdere kö., 29.viii.1995, G. Tümen (ESSE 12250)! Muğla, 23.vii.1949, T. Baytop (ISTE 2429, 2435)! Marmaris'e 30 km, yol kenarı, 320 m, 20.v1.1980, N. ve E. Özhatay, E. Tuzlacı (ISTE 44877)! Sandras da., Akköprü orman işletmesi-Armutveren, 700 m, N. ve E. Özhatay (ISTE 44932)! Fethiye, Kemer yakını, ulualan ağaçlandırma sahası, 400 m, 1.vıı.1983, E. Tuzlacı (ISTE 51462)! Fethiye, Baba da., batı etekleri, Gıdırak, Belceğiz arasındaki yamaçlar, 30 m, E. Tuzlacı (ISTE 53130)! Köyceğiz, Sandras da., Ağla sırtları, orman altı, 900 m, 22.vı.1980, N. ve E. Özhatay (MUFE 4302)! C3 Antalya: Antalya, 140 m, 14.v.1971, R. Çetik 3760 (ANK)! Maki zonu, 2.iv.1984, Y. Akman 13638 (ANK)! Antalya'nın 12 km güneybatısı, Karadağ, 40 m, 22.1v.1954, Nijhoffet et all. 612 (ANK)! Antalya, Plaj ve kayalar, H. Birand 17 (ANK)! Alanya-Manavgat, 23.v1.1977, İ. Akbulut (IZEF 2178)! Kemer, Faselis koyu ve cevresi, 0-150 m, 23.v1.1978, H. Pesmen 4036, B. Yıldız, Ş. Kaplan (HUB 22690)! Kemer, Beldibi köyü üstü, kalkerli derin vadi, 30-100 m, 20.v11.1978, H. Pesmen 3897, A. Güner (HUB 22687)! Side cevresi, 5.v1.1970, A. Pamukçuoğlu, Quezel (HUB 22688)! Kemer yolu, karayolu tüneli çevresi, masif kalker kayalığı, 23.11.1978, H. Peşmen 3617, B. Yıldız (HUB 22689)! Antalya-Kurşunlu şelalesi, Barış parkı, 2.vı.1989, K.H.C. Başer, N. Kurtar (ESSE 8764)! Antalya-Muğla karayolu, Tekirova mevkii, 1.v1.1989, K.H.C. Başer, N. Kurtar (ESSE 8763)! Antalya-Kemer karayolu 15 km, 14.1v.1991, K.H.C. Başer, N. Öztürk (ESSE 9378)! Antalya-Konaklı kö., 1991, K.H.C. Başer (ESSE 9697)! Alanya, Konaklı kö., vın. 1991, G. Tümen (ESSE 9735)! ibid., 19.1x. 1992, G. Tümen (ESSE 10497)! ibid., 20.vtn.1994, K.H.C. Başer (ESSE 11857)! Hisarçanlı TV kulesi, 21.vtn.1989, K.H.C. Başer, N. Kurtar (ESSE 10123)! Alanya, Altes, 18.1x.1995, K.H.C. Başer (ESSE 11038)! Manavgat-Alanya, Çavuşköyü

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Satureja cuneifolia Ten.

A1(E) Canakkale: Lapseki, Ecialan kö. civarı, 22.viii.1995, G. Tümen (ESSE 12029)! Gökçeada, Araz tepesi doğusu, ca. 300 m, 28.vii.1975, (EGE 15921)! Gökçeada, Demirkaya te., ca. 672 m, 10.viii.1976, (EGE 15923)! Gökçeada, Ulukaya te., ca. 630 m, 21.v.1975, (EGE 15922)! Gökçeada, Rıhı tepesi doğusu, ca. 350 m, 11.viii.1976, (EGE 15924)! A2 Bilecik: Dingler, (ANK)! A4 Zonguldak: Safranbolu, 29.viii.1995, G. Tümen (ESSE 11675)! A5 Amasya: kayalıklar, 1.viii.1994, N. Ermin (ESSE 10666)! B1 Balıkesir: Balıkesir-Kazdağ, Babadağ yöresi, Kapıkule zirve, 17.vn.1991, G. Tümen, H. Çakır (ESSE 10440)! Ayvalık yolu üzerindeki denize bakan yamaçlar, 10.v11.1989, G. Tümen (ESSE 8461)! Edremit, Ayvacık yolu üzerinde, Küçükkuyu'ya 10 km, 100-150 m, 5.viii.1989, G. Tümen (ISTE 62546)! Canakkale: Balikesir-Canakkale, Avvacik'a varmadan, karayolları dinlenme tesisi, 50-100 m, kayalık yamaçlar, 28.1x.1995, G. Tümen (ESSE 12021)! Küçükkuyu, Subaşı-Ayı kayası mevkii, 24.1x.2001, N. Sadıkoğlu (ISTE 81653)! İzmir: Ödemiş, Bozdağ, 1050 m, 24.11.1962, K. Karamanoğlu 890 (ANK)! Ödemiş-Bozdağ, 5.x.1992, G. Tümen (ESSE 10112)! Ödemiş, Birgi, G. Tümen (ESSE 10081)! Ödemiş, Bozdağ, vı.1993, G. Tümen (ESSE 10203)! ibid., 1991, G. Tümen (ESSE 9559)! ibid., v11.1994, G. Tümen (ESSE 10992)! ibid., 20.v11.1995, G. Tümen (ESSE 11670)! Bozdağ, Büyükçavdar Y. sırtları, 1400 m, 9.x.1980, A. Baytop (ISTE 45905)! Manisa: Alaşehir-Bozdağ, Azıtepe kö., 20.v111.1993, G. Tümen (ESSE 10170)! B2 İzmir: Bozdağ, Yayla, ca. 1100 m, 05.x.1994, E. Saver, U. Zeybek (IZEF 3058)! Kiraz, Küçük Menderes kenarı, x.1992, G. Tümen (ESSE 9867)! Kiraz çevresi, 25.1x.1995, G. Tümen (ESSE 11656)! Kiraz-Kücükmenderes ırmağı cevresi, 320 m, 21.ix.1995, F. Yılmaz (ESSE 11672)! Ödemis, Gölcük, 21.x.1980, E. Tuzlacı, N. Sütlüpınar, A. Mericli, Y. Kalav (ISTE 45944)! Denizli: Civril-Akdağ, 6.x.1983, (EGE 25943)! Manisa: Alaşehir, Azıtepe kö., Bozdağ, 20.1x.1994 (ESSE 10954)! Kula, Azıtepe, 20.v11.1995, İ. Cınar, G. Tümen (ESSE 12129)! Salihli, Bozdağ, 2.x.1997, G. Tümen (ESSE 12474)! B3 Eskişehir: Eskişehir: Seyitgazi yolu 10-15 km, sol taraftaki araziden 1 km, 1972, K.H.C. Başer, (ESSE 179)! Kanlıpınar göleti, kayalıklar, 6.1x.1991, K.H.C. Başer (ESSE 9591)! Mayıslar-Eskişehir 2 km, 12.vı.1991, K.H.C. Başer, A. Kaya (ESSE 8452)! Dağküplü-Mayıslar, 9 km, 14.vı.1994, K.H.C. Başer, A. Kaya (ESSE 11147)! Mayıslar yakını, vııı.1992, G. Tümen (ESSE 10031)! Mayıslar-Dağküplü 1 km, 28.v.1994, G. Tümen (ESSE 10481)! Şöförler çeşmesinden sonra 8. km, kayalıklar, 20.vı11.1991, A. Kaya, N. Ermin, T. Özek (ESSE 9200)! Gökçekaya barajı, v111.1994 (ESSE 10726)! Gökçekaya barajı, Hidrofilik santralinin çevresi, Doğu Savak bölgesi, 25.1x.1998, N. Tabanca, A. Altıntas (ESSE 12754)! Kanlıpınar göleti, 6.ix.1991, K.H.C. Başer, A. Kaya (ISTE 63602)! C1 Muğla: Bodrum, Bitez kö. cevresi, d.s. 6.x.2002, E. Tuzlacı (MARE 8545)! ibid., bahçelerin üzerindeki sırtlarda, 13.xı.1976, E. Tuzlacı (ISTE 36290)! C2 Antalya: Altes bahçesinden, vın.1995, N. Mumcuoğlu (ESSE 11973)! Aydın: Karacasu, Türer Tarım Orman Ürünleri İthalat İhracat Sanayi, 30.vm.2000 (ESSE 13287)! Denizli: Bozdağ, 14.7.1947, P.H. Davis 13333 (ANK)! Babadağ-Başaran yaylası, 1200 m, 7.1x.1994, (EGE 18904)! Babadağ, 900-1000 m, 23.v111.1950, (EGE 27252)! Babadağ yaylasından kooperatif evlerinin başlangıcı, 26.1x.1997, G. Tümen (ESSE 12484)! Olukbaşı-Geyran Y., Koyunini de. üstü, 1540 m, 11.vn.1999, K.H.C. Baser 1618, H. Duman, A. Altıntaş (ESSE 12859)! Babadağ, Türer Tarım Orman Ürünleri İthalat İhracat Sanayi, 30.viii.2000 (ESSE 13290)! Babadağ, 900-1000 m, 23.viii.1950, P.H. Davis 18425 (ISTE 52321)! Muğla: Fethiye, Kemer, Çayan kö., Gavır mevkii, x.1993, G. Tümen (ESSE 10466)! Ortaca, Cövenli Y., dağlık yerler, viii.1997, G. Tümen (ESSE 12790)! C3 Antalya: Kemer, 2000-2200 m, 10.v11.1947, P.H. Davis 14190 (ANK)! Kemer, Tahtalı dağ, Çukur Y.-Ağla Y., kalkerli kuzev yamac, Cedrus libani ormanı, 1100-1650 m, 23.vui.1978, H. Pesmen-A. Güner, P.H. Davis 4074 (ANK)! ibid., H. Pesmen 4100, A. Güner (HUB 22657)! ibid., H. Pesmen 4074, A. Güner (HUB 22660)! Elmalı civarı, Kayalar, 1968, Quézel et all. 29 (ANK)! Elmalı, Beydağ, 2000 m, 28.vıi.1960, Khan et all. 288 (ANK)! Göynük, 50 m, 6.vii.1949, P.H. Davis 15026 (ANK)! Han Boğazı, 2.ix.1947, P.H. Davis 14722 (ANK)! Tahtalı Da., W Yukarı Beycik, 2000-2350, 1.vıı.1984, (EGE 31175)! Çalbalı Da., SE Bakırlı Dağ, 1550-1650 m, 18.vii.1984, (EGE 31174)! Kemer, Beycik kö., Tahtali Da., ca. 1250-2100 m, 26.vii.1980, (EGE 32352)! Kemer, Yaylabuz dere-Çukur Y. Cedrus libani ormanı, 1200-1500 m, 28.vu.1979, H. Peşmen 4448, A. Güner, Ş. Kaplan

(HUB 22661, ISTE 52634)! Kemer, Beycik kö. üstü, kalkerli kayalık arazi, P. brutia-C. libani ormanı ve alpinik step, 800-1900 m, 19.v1.1978, H. Peşmen 3891, A. Güner (HUB 22652)! Antalya-Burdur karayolu, 17.1x.1994, K.H.C. Başer (ESSE 11035)! Akseki, Sadıklar kö., 14.vn.1974, A. Alpaslan (ISTE 30778)! Çalbalı da., Fesleğen Y. yakını, 1850 m, 4.viii.1980, N. Özhatay, E. Tuzlacı, B. Çubukçu, A. Meriçli (ISTE 45631)! Burdur: 11.v1.1993, G. Tümen (ESSE 10082)! Isparta: Eğridir, Barla Dağı, 2000 m, 1.v11.1960, Khan et all. 396 (ANK)! Ş. Karaağaç, Kızıldağ Milli Parkı, Kızıldağ kuzey yamacı, 1300 m, 24.vıı.1994, B. Mutlu 1078 (HUB 22662)! Eğirdir, Anamas, Yaka kö., Kapız de., kalkerli sap ve derin vadi, 1250-1450 m, 5.viii.1974, H. Peşmen, A. Güner 1881 (HUB 22653)! Sütçüler, Çandır-Akçay, yangın gözetleme kulesi civarı, vıı.1995, K.H.C. Başer, H. Duman, A. Altıntaş (ESSE 11830)! Sütçüler, Çandır, 1800 m, ıx.1999, S. Oflaz (ESSE 13357)! Sütçüler, Darıbükü, N37°34' E31°11', 12.vni.2001, N. Sadıkoğlu (ISTE 81648)! C4 Antalya: Alanya, Gönül dere, 1000 m, K. Karamanoğlu, 27.yın.1947, P.H. Davis 14293 (ANK)! Gazipasa, Sugözü kö., Maha Y., 450 m, 5.yın.1983, H. Sümbül 2382 (HUB 22655)! Alanva, Köprübası mevkii, Arpalık Y., kuzev sırtları, 1600-2000 m. 10.viii.1994, H. Duman (ESSE 10731)! Alanya, Gökbel yolu, Bucak Cökelek Y., Alanya'dan 32. km, 1330 m, 17.vii.1995, K.H.C. Baser 1172, H. Duman, A. Altıntaş (ESSE 11506)! Karaman: Ermenek, 1300 m, 10.vui.1947, P.H. Davis 16156 (ANK)! Ermenek, Göktepe nahiyesi, Dumlugöze (Muzvadi) kö., Aşakbel tepesi, 1900 m, 13.1x.1983, H. Sümbül 2405 (HUB 22656)! Ermenek, 1300-1400 m, 13.vin.1949, P.H. Davis 16156 (ISTE 52320)! Ermenek Yellibel yolu, Tekecatı, yol kenarı, 26.vn.1991, F. ve A.H. Mericli (ISTE 63770)! Konya: Bozkır, Dikilitaş Y., R. Çetik-T. Ekim-E. Yurdakulol, 6.vui.1967, Huber-Morath 12 (ANK)! Bozkir, 1000 m, 7.ix.1969, P.H. Davis 16610 (ANK)! Bozkır, Yılanlı kaya mevkii, kaya üzeri, 1550 m, 11.vıı.1989, H. Sümbül 3389 (HUB 22650)! Mersin: Anamur, Hamitseydi Boğazı, Ermenek, 1500-1700 m, P.H. Davis 16241 (ANK)! Gülnar-Silifke, 1000 m, 28.1x.1994, kavalıklar, M. Vural 7236, M. Kovuncu, M. Ekici (HUB 24382)! Gülnar cevresinden, v11.1991, G. Tümen (ESSE 9558)! Silifke-Ardıçkuyusu Y., vın. 1992, G. Tümen (ESSE 10027)! Gülnar çevresi, 20.vın. 1995, G. Tümen (ESSE 12116)! Silifke, Sarıaydın kö., 1750 m, 4.v.1995, T. Baytop (ISTE 71374)! C5 Adana: Dildil Dağ, Başkonur Y., Hüseyin oluk çeşmesi, 1800 m, 27.vui.1949, P.H. Davis 16400 (ANK)! Bahçe (Amanus), Dildil dağ between Başkonus Y. and Hüseyin Oluk Y., 1800 m, 27.vn.1949, P.H. Davis 16400 (ISTE 52319)! Konya: Ereğli, Aydos Da., Delimahmutlu-Çakıllar, meşe ormanı, kalker anakaya, 1600 m, 29.vı11.1977, S. Erik 2652 (HUB)! Ereğli, Aydos Da., Delimahmutlu, otlak tepe, kalker anakaya, bozkır, 1600 m, 19.vui.1978, S. Erik 3054 (HUB 22658)! Ereğli, Aydos Da., Kayasaray, mermer anakaya, 1700 m, 15.vii.1977, S. Erik 2589 (HUB 22659)! Halkapınar-İvriz, Karacaser dağı yamacı, 1200 m, 13.vu.1998, Y. Yaman (MUFE 115)! Mersin: Mersin-Aslanköy Y., vui.1990, G. Tümen (ESSE 9557)! Silifke-Sulama Y., Karadedeli kö., G. Tümen (ESSE 10174)! Erdemli, Büyüksorgun Ziyaret da., 950 m, yamaclar, 8.vı.1996, G. Tümen (ESSE 12249)! Bolkar da., Aslanköy, Kazangöl pınarı, kayaların üstünde, 1600 m, 12.vın.1976, K. Alpınar (ISTE 35832)! C6 Hatay: Altınözü, Yanıkpınar kö., Kayacık mevkii, kayaların üzerinden, 22.x.1995, B. İshakoğlu (ESSE 12028)! Hassa, taslık arazi, 21.x.1996, G. Tümen (ESSE 12271)!

Thymbra capitata (L.) Cav.

A1(E) Çanakkale: Eceabat-Abide yolu, 17.v1.1986, N. Zeybek (IZEF 2284)! Bozcaada, Sulubahçe güneyi, ca. 25 m, 7.viii.1976, (EGE 20491)! Bozcaada, Göztepe batisi, çayır, ca. 80 m, 14.vi.1976, (EGE 20487)! Eceabat, Kabatepe sırtları, 50 m, 24.1x.1984, A. Çırpıcı, T. Ekim, H. Malyer (ESSE 7516)! Eceabat, Kilitbahir, Havuzlar, Şarlayandere yolu, 14 m, 24.1x.1984, A. Çırpıcı, H. Malyer (ESSE 8044)! Eceabat-Kabatepe mevkii, Saroz körfezi, yamaçlar, 16.v11.1991, G. Tümen (ESSE 10009)! Gelibolu-Burhanlı köyünün hemen üstü, denize bakan yamaçlar, 29.viii.1991, G. Tümen (ESSE 10023)! Eceabat-Gelibolu yolu, 10.vii.1960, A. Baytop (ISTE 6091)! Gelibolu-Eceabat, Gelibolu'dan 25 km ileride, Galata ovası, sırtlarda, 29.vn.1971, A. Baytop (ISTE 20729)! İmroz adası, Dereköy, 19.vn. 1974, Y. Saviç (ISTE 30519)! Eceabat, Şeddülbahir civarı, 21.vn.1975, N. ve E. Özhatay (ISTE 32999)! Tekirdağ: Gelibolu, Buharlı köyü 10 km sonra, maki, K. Karamanoğlu, 617 (ANK)! B1 Balıkesir: Marmara adası, vı11.1991, G. Tümen (ESSE 9818)! Marmara adası-Çınarlı kö., koylardan biri, 25.viii.1991, G. Tümen (ESSE 10219)! Canakkale: Bozcaada, bati burnu, adanın en bati ucu, c. 30 m, açık kireçli kayalıkların bulunduğu tepeler, 28.vı11.1994, A.J. Byfield, S. Atay B 1271 (ISTE 67526)! İzmir: Urla-Kalabak, 25.v111.1977, İ. Akbulut (IZEF 339)! Urla İskelesi, 12.v1.1977, İ. Akbulut (IZEF 357)! Bornova, Sabuncubeli, 150 m, 11.vii.1982, N. Zeybek (IZEF 665)! Bornova, askeriye çıkışı, 24.vi.1977, İ. Akbulut (IZEF 2176)! Bornova-Papaz tepesi, 20.x.1995, E. Saver (IZEF 3638)! Hills NE of Bornova, 1.vii.1969, (EGE 8094)! Eski İzmir'in cıkısı, çatalkayaya doğru, ca. 250 m, 28.vn.1982, (EGE 26499)! Çeşme, 1.x.1967, (EGE 4614)! Çeşme, Çiftlik kö., 10.v1.1971, (EGE 7612)! Karaburun, Saip dağ, 4.v11.1965, (EGE 6052)! Mordoğan, Ziraat Fak. Kampı üstü, kalkerli toprak, c. 100 m, 6.vui.1978, Ş. Yıldırımlı 1055 (HUB 22961)! Çeşme, Askeri lojmanların arkası, 100 m, 28.vii.1997, B. Mutlu 1917 (HUB 22964)! Mordoğan, 28.vii.1995, S., M. ve H. Alan (ESSE 11845)! Çeşme, Dalyan, 24.v11.1997, G. Tümen (ESSE 12810)! Seferihisar, 11.x.2000, (ESSE 13298)! Çeşme, Çiftlik kö. ilerisi, deniz kenarı, 1.vıı.1969, T. Avcıgil (ISTE 15929)! C1 Aydın: Kuşadası-Aydın yolu, İ. 18.vı.1977, İ. Akbulut (IZEF 381)! Kuşadası, 21.vı.1964, (EGE 2328!, 2332!, 8721)! Didim-Söke, c. 10 m, 5.vıı.1984, B. Dinçtürk 1006 (HUB 22950)! Germencik, 10.v11.1993, G. Tümen (ESSE 10098)! Kuşadası, 25.v1.1996, G. Tümen (ESSE 12182)! İzmir: Selçuk, Pamucak beli, 27.vn.1965, (EGE 6060)! Selçuk, Meryemana, 15.vn.1966, (EGE 5349)! Muğla: Datça, Karaköy civarı, Maki açıklıklarında, 200 m, 14.vın.1985, M. Demirörs 2058 (ANK)! Datça, 19.vn.1996, (EGE 5342)! Datca, (EGE 5978)! Bodrum, Geriş kö. güneyindeki yamaçlar, 250 m, 14.v1.2001, E. Tuzlacı (MARE 6793)! Bodrum, Bitez kö. kuzevindeki dağ yamacları, 100 m, 15.v1.2001, E. Tuzlacı (MARE 6813)! Bodrum: Bodrum-Çamlık kö., 280 m, 5.v.2002, E. Tuzlacı (MARE 7536)! Datça, Gebekum, 6.ix.2001, E. Tuzlacı (MARE 7157)! Bodrum, Kadı kalesi çevresi, 20.v11.1994, G. Tümen (ESSE 10975)! Datça, Bozdağ (Kocadağ), Mesudiye kö. üstleri, 600 m, 3.vn.1983, E. Tuzlacı (ISTE 51503)! Datça-Marmaris, Datça'dan 13 km, Gebekum mevkii, kumullarda, d.s., 5.vii.1983, E. Tuzlacı (ISTE 51590)! Bodrum, Bitez kö. çevresi, 30 m, 7.v11.1983, E. Tuzlacı (ISTE 51609)! Datça, Kocadağ kuzey etekleri, Karaköy yakını, d.s., 12.v.1984, E. Tuzlacı (ISTE 53290)! Bodrum, Gölköy, Eski Gölköy Lisesi sırtları, 11.vı.2003, N. Sadıkoğlu (ISTE 81672)! C2 Antalya: Elmali-Akdağ volu, 18, vi. 1968, A. Pamukcuoğlu, Quezel (HUB 22962)! Kas-Fethiye karayolu 32. km, 150 m, vol kenari, yamaclar, 20.vi.1995, K.H.C. Baser 1039, H. Duman, A. Altintas (ESSE 11359)! Muğla: Marmaris-Datca, 17.vu.1960, Khan et all, 79 (ANK)! Köyceğiz, 03.vu.1977, İ. Akbulut (IZEF 2171)! Marmaris, Hisarönü kö., 10.x1.1984, N.Zevbek (IZEF 2912)! Köyceğiz, Dalyan, Sülüngür gölü, 27.v1.1981, (EGE 32468)! Fethive-Tersakan, 9.vii.1966, (EGE 5443)! ibid., (EGE 5981)! Marmaris, Pamucum orman dinlenme kampi vakını, yamaclar, G. Tümen (ESSE 9816)! Marmaris-Kadıkale, vu. 1993, G. Tümen (ESSE 10202)! Marmaris'e 30 km, 320 m, vol kenari, 20.vi.1980, E. Tuzlaci, N. ve E. Özhatay (ISTE 44877a)! C3 Antalya: Kemer-Göynük, 5 km, kumul, 29.v11.1980, H. Peşmen 4915 (HUB 22963)! Beycik kö., İkiağızlar mevki, 31.v.1990, K.H.C. Başer (ESSE 8786)! Konyaaltı, 10 m, kumullar, N. Özhatay, E. Tuzlacı, B. Çubukçu, A. Meriçli (ISTE 45665a)! C6 Hatay: İskenderun, Soğukoluk, Maki çalılığı boşluklarında, 25.vı.1944, B. Kasaplıgil 69 Kew (ANK)! İskenderun, Soğukoluk, Maki zonu, 350 m, 25.vı.1967, Y. Akman 7722 (ANK)!

Thymbra spicata L. subsp. spicata

A1(A) Balıkesir: Gönen, Taşocağı sırtları, 40 m, 30.vı.1996, P. Eryaşar (MARE 5107)! Canakkale: Ezine, Pazarköy, 150 m, 25.v1.2002, G. Emre (MARE 8392)! Ezine, Akköy, 170 m, 13.1v.2002, G. Emre (MARE 8253)! A1(E) Çanakkale: Gelibolu-Eccabat, Eccabat'a 10 km kala, 9.x1.1968, A. Baytop, B. Çubukçu (ISTE 14713)! Havuzlu-Behramlı, 9.xı.1968, A. Baytop, B. Çubukçu (ISTE 14732)! Tekirdağ: Kumbağ sahil yolu, 13.vın.1983, N. Zeybek (IZEF 2972)! Tekirdağ-Barbaros, 25.111.1968, A. Baytop, G. Atila (ISTE 12432)! Ganosdağ, Gaziköye inerken, 100 m, 14.v11.1968, A. Baytop (ISTE 13577a)! Keşan-Malkara, il hududu, 1.v111.1971, A. Baytop (ISTE 20858)! Tekirdağ-Kumbağ yolu, Barbaros'a 1 km kala, çeşme üstündeki sırtlarda, 16.vıı.1974, N. ve E. Özhatay (ISTE 30432)! Karıştıran-Şarköy, Çınarlıdere kö. ayrımı, 19.vıı.1992, E. Akalın (ISTE 64511)! Merkez, Dedecik kö. çıkışı, yol kenarları, 30.vı.1994, E. Akalın (ISTE 67409)! A2(A) Bursa: İznik yol ayırımı, maki, 50 m, E. Yurdakulol-M. Kılınç-M. Aydoğdu (ANK)! Bursa'nın 25 km kuzeyi, Mudanya'nın 5 km batısı, ca. 20 m, 21.vi.1973, (EGE 15463)! Mudanya'nın 5 km batısı, Mudanya-Zevtinbağı (Tirilye), 20 m, 21.vi.1973, (EGE 23422)! İstanbul: Sile, Akcakese kö., 80 m, 30.v.1996, E.T. Fenercioğlu (MARE 4835)! ibid., 5.v1.1995, E.T. Fenercioğlu (MARE 4677)! Şile, İmrenli kö., 30 m, 8.vı.1995, E.T. Fenercioğlu (MARE 4665)! A2(E) İstanbul: Kücükcekmece gölünün kuzey ucu-Hosdere, kuru tepeler, 17.v11.1969, A. Baytop (ISTE 15686)! Büyük Halkalı köyünden dönüste, 3.vii.1970, N. Özocak, E. Özhatay (ISTE 18148)! Halkalı batısındaki kirecli yamaclar, 5.v11.1976, E. Tuzlacı (ISTE 40293)! İzmir'den getirilen ve Maltepe'de bahçede yetiştirilen numunelerden, 25.v1.1967, A. ve T. Baytop (ISTE 12060)! A3 Adapazari: Pamukova'dan Sapanca'ya doğru dört yol ayrımı, ca. 40 m, 31.vn.1984 (EGE 17883)! Geyve-İznik, Osmaneli yol ayrımından 5-6 km, Mekece kö.-İznik, 20.v1.190, E. Tuzlacı, M. Öksüz, F. Hırlak (MARE 2607)! Bolu: Abant gö. civarı, vii.1956, Ş. Tercan (ISTE 4637)! A4 Karabük: Kum ocakları karşısı, 700 m, 21.vı.1985, M. Demirörs 1558 (ANK)! Kastamonu: Araç, P.nigra, 900 m, 24.vu.1981, M. Demirörs 414 (ANK)! A5 Amasya: Amasya civarı, 13.vu.1956, T. Baytop (ISTE 4648)! A6 Tokat: Reşadiye, 10 km batısı, Kelkit vadisi, Taşlı yerler, 10.vn.1955, R. Çetik 1047/548 (ANK)! B1 Balıkesir: Akçay-Edremit, 1x.1986, K. Sır (MARE 678)! Çanakkale: Çanakkale-Ezine'nin 4 km kuzeyi, 8.v111.78, 160 m, (VANF 664)!, İzmir: İzmir, 8.vı.1931 Krause 337 (ANK)! 12.vı.1935, Gassner 70 (ANK)! Çeşme, 16.vı.1976, F. Şekerci (IZEF 323)! Urla İskelesi, 12.vı.1977, İ. Akbulut (IZEF 327)! Gümüldür-Karacadağ, Değirmendere, ca. 100 m, 16.v1.1986, (EGE 21792)! Bornova, Hacılar kırı, Naldöken'in kuzeydoğusu, 25 m, 1.v1.1967, (EGE 6369)! Hills NE of Bornova, 1.v1.1969, (EGE 8089)! ibid., 26.v.1969, (EGE 4210)! Ödemis, Gölcük-Bozdağ, 4.vii.1966, (EGE 6033)! İnciralti-Urla, 20.ix.1962, (EGE 873)! Selçuk, Meryemana, 18.vi.1971, A. Pamukçuoğlu (HUB 22971)! Kemalpaşa, piknik yeri, Nif D. Etekleri, 280 m, 23.v1.1990, E. Tuzlacı (MARE 2747)! İzmir Çeşme yolu, İçmeler'e varmadan, 21.11.1967, A. ve T. Baytop (ISTE 10690)! Manisa: Bozdağ, 16.v11.1984, S. Yıldırımlı 7486 (HUB 22972)! Akhisar, Yel değirmeni, H. Bağda 460 (ANK)! Akhisar, Yel değirmeni, 9.vı.1942, H. Bağda (ISTE 1362)! Manisa da., 3.v.1961, T. Baytop (ISTE 6415)! B6 Adana: Saimbeyli, Himmetli köyü güneyi, Mermer kayalık, 900 m, 20.vı.1978, T. Ekim 3513 (ANK)! Sasak, Hacin, Manissadjian 1012 (ANK)! B8 Diyarbakır: Hani-Dicle, Hani'den 14 km, 900 m, kıraç yerler, killi topraklar, 11.vı.1979, E. Tuzlacı, M. Saraçoğlu (ISTE 42188)! Siirt: Silvan-Kurtalan, 1000 m, 24.v1.1954, Davis 22176 (ANK)! B8/9 Siirt: Sirvan, Cevizlik Siirt yol ayrımı çevresi, bozuk meşelik, kuru yamaçlar, kalkerli arazi, 1000 m, 13.1v.1980, A. Güner 2349, M. Koyuncu (HUB 22974)! C1 Aydın: Kuşadası, 250 m, 18.v1.1977, İ. Akbulut (IZEF 341)! Kuşadası, 01.1.1950, M. Polat (IZEF 1759)! Söke-Kuşadası, ca. 300 m, 21.v1.1977, İ. Akbulut (IZEF 2165)! Kuşadası, 7.x1.1967, (EGE 4692)!

C2 Antalya: Kaş, Elmalı, 27.vii.1960, Khan et all 226 (ANK)! Fethiye-Kemer, 02.vii.1977, (IZEF 344)! Kumluca, Mavikent, Yenice kö., 8.1v.1998, B. Sadak (MARE 5813)! Aydın: Sultanhisar, 01.1.1950, M. Polat (IZEF 1769)! Denizli: Tavas'dan 10 km sonra, Kızılcam acıklıklarında, 9.vu.1962, 840 (ANK)! Tavas-Denizli, Tavas'tan 7 km, yol kenarındaki killi yamaçlar, 1100 m, 7.vn.1989, E. Tuzlacı (MARE 2019)! Denizli, 7.vn.1905, St. Lager (ISTE 1361)! Honaz da., Karatepe kö. üstü, 730 m, orman yolu kenarı, 1.vıı.1972, E. Tuzlacı (ISTE 22899)! Denizli Kazıkbeli yolu, Karatepe Gökkaya ağaçlandırma sahası, 750 m, 7.vı.1973, A. Baytop, E. Tuzlacı (ISTE 25505)! Honaz'ın 4-5 km kuzeydoğusu, 500 m, dere kenarı, 13.vı.1973, E. Tuzlacı (ISTE 25917)! Denizli kuzeyi 4 km kala (Aydın yönünden), 8.v11.1973, E. Tuzlacı (ISTE 26267)! Muğla: Fethiye, S of Kestep, Kaş, 11.v1.1969, (EGE 7815)! Fethiye, Kemerbucağı, Dereçatı kö., Bayaslar mevkii, c. 300 m, 15.v1.1967, (EGE 5800)! Fethiye, Bayaslar, Dereçatı, 15.vi.1967, (EGE 4914)! Muğla, 23.viii. 1949, T. Baytop 1363 (HUB 22976)! Köyceğiz çevresi, 60 m, yol kenarı, 22.v.1991, A. Güner 9250, M. Vural, H. Şağban (HUB 2966)! Köyceğiz, Candır kö., 360 m, kızılçam ormanı, 19.vı.1991, A. Güner 9516, M. Vural, H. Duman, A. Dönmez, H. Şağban (HUB 22965)! Muğla, 23.vm.1949, T. Baytop (ISTE 1363)! Fethiye, Baba da. volu, Fethiye'ye 4 km, kaya üstü, 120 m, 1.vii. 1983, E. Tuzlaci (ISTE 51428)! Fethive-Kemer, Kemer vakini, 100 m, 1.vii. 1983, E. Tuzlaci (ISTE 51430)! Muğla-Tavas, 2-3 km from Muğla, 150 m, D. 35506 (ISTE 51819)! C3 Afvon: Basmakci, 27.vi.1995, U. Zeybek (IZEF 3417)! Antalya: Alanya, Davis 14486 (ANK)! Göynük, 50 m, Davis 15032 (ANK)! Kemer, Kumluca, marnlı topraklar, 250 m, 26.v.1984, Y. Akman 13643 (ANK)! Lara, 29.v1.1938, Gassner 1173 (ANK)! Kargı çayı, 24.vui.1947, Davis 14417 (ANK)! Alanya, Alarahan, 23.iv.1978 (EGE 15295)! Alanya'nın 5 km batısı, 80 m, 17.vı.1966, (EGE 6379)! Side çevresi, 5.vı.1970, A. Pamukçuoğlu-Quezel (HUB 22970)! Akseki, 18.vn.1967, H. Küçük (ISTE 12006)! Çakırlar, Akdamlar kö. üstleri, Fesleğen Y. yolu, 200 m, 4.vn.1980, N. Özhatay, E. Tuzlacı, B. Çubukçu, A. Meriçli (ISTE 45662)! Manavgat üstleri, Beşkonak-Olukköprü, 100 m, 3.vii.1982, G. Çakırer, A. Öztekin (ISTE 49125)! Manavgat'tan 21 km ileride, Oymapınar barajı civarı, 100 m, 19.v1.1983, H. ve G. Çakırer (ISTE 51086)! Finike-Elmalı yolu, Arif köyüne gelmeden Aygır suyu mevkii, Arycandra Lokantası civarı yamaçlar, 13.v11.1990, A.H. ve F. Meriçli (ISTE 62458)! Isparta: Sütçüler, Darıbükü, N37°34' E31°11', 12.vı11.2001, N. Sadıkoğlu (ISTE 81647)! C4 Antalya: Gazipaşa, Sugözü kö., P. brutia ormanı, 1600 m, 12.vıı.1982, H. Sümbül 2983 (HUB 22967)! Kumluca, Andrasan koyu kuzeyi, serpentin arazi, P. brutia ormanı, 0-120 m, 8.vı.1979, H. Peşmen 4384 et Güner (HUB 22969)! Alanya-Gazipaşa 35. km, 9.viii.1965, N. ve M. Tanker, E. Sezik (ISTE 8280)! Alanya, Emirgan civari, yol kenarlari, 21.viii.1965, E. Sezik, M. Tanker (ISTE 8316)! Alanya, Deretürkenaz yolu, Bektas çeşmesi, 325 m, 24.11.1966, A. ve T. Baytop, N. Tanker, E. Sezik (ISTE 8529)! Alanya kalesi, 26.v.1966, A. ve T. Baytop, B. Cubukcu (ISTE 9670)! Güzelbağ-Gündoğmus, 520 m, 5.viii. 1980, E. Tuzlaci, B. Cubukcu, A. Mericli (ISTE 45675)! Karaman: Ermenek, Kazancı kasabası civarı, 650-850 m, 21.vı.1984, H. Sümbül 2983 (ANK)! Ermenek-Kazancı kasabası civarı, 650-850 m, 21.v1.1984, H. Sümbül 2983 (HUB 22968)! Bucakkısla, Kuru dere icleri, 450 m, 30.v.1979, M. Vural 1795 (ANK)! Mersin: Mut-Silifke, Göksu nehrinin vadisi, vadinin sırtlarında kumlu topraklar, vı.1994, A.J. Byfield B 1212 (ISTE 67467)! Anamur-Silifke, Anamur'un 40 km doğusu, 210 m, 23.1v.1985, M. Nydegger (ISTE 74762)! C5 Adana: Karsantı, Ortaca köprüsü civarı, 1000 m, 26.vı.1973, E. Yurdakulol 1542 (ANK)! Süphan de., Belen köy, 900-1000 m, 2.vii.1952, Davis 19570 (ANK)! Feke Bakırdağ arasındaki orman yolu, Feke'den 15 km (Feke'nin kuzeybatısı), 820 m, maki arası, 31.vıi.1979, E. Tuzlacı, M. Saraçoğlu (ISTE 43388)! Kozan, Akçalı Orman İşletmesi, istif sahası civarı, 16.vıı.1985, İ. Saracoğlu, T. Ersöz 3010 (ISTE 55628)! Hatay: Tekepınar aşağısı, Musa da. etekleri, 330 m, maki arası, 17.vı.1979, E. Tuzlacı, M. Saraçoğlu (ISTE 42305)! Mersin: Bahçe, Paslikale, P.brutia ormani, 450 m, 22.vi.1972, T. Uslu (ANK)! Mersin Gözne yolu, Buluklu kö. üstü, 420 m, 5.v1.1981, E. Tuzlacı (ISTE 46365)! Gülnar-Aydıncık, Gülnar'a 15 km, Ezkiyörük kö. çevresi, 850 m, 15.v1.1981, E. Tuzlacı (ISTE 46509)! Niğde: Ulukışla, Aydos Da., Ali hoca yakınları, kalkerli yamaç, 1200 m, 28.vu.1977, S. Erik 2989 (HUB 22973)! C6 Gaziantep: Besni, v.1936, Gleisberg 110 (ANK)! Hatay: Hatay-Hassa, L. Behçet s.n. (VANF)! Harbiye'nin güneyi, kurak, 18.x.97, 100-130 m, L. Behçet 9 (VANF)! Antakya, Habibineccar da., Taş ocağı yanı, 24.vı.1974, A.H. Meriçli (ISTE 29970)! Altınözü, 3.vu.1975, M. Miski (ISTE 32979)! Belen sırtları, 20.vın.1976, M. Miski (ISTE 35959)! Reyhanlı, Harran (Kavalcık) kö., Suriye sınırı bölgesindeki ekilmemiş kıraç bölgeler, 26.v.1977, E. Tuzlacı (ISTE 37181)! Antakya yakınları, 25.vı.1982, M. Miski (ISTE 49934)! İskenderun, pazardan satın alındı, 6.1v.1989, T. Baytop (ISTE 60150)! Antakya, Habibneccar da., 28.v1.2004, N. Sadıkoğlu (ISTE 81696)! Yayladağ, Sebenoba kö, tarla kenarı, 29.v1.2004, N. Sadıkoğlu (ISTE 81648)! Maras: Maras Göksun volu, Ceyhan köprüsü cevresi, 700 m, 13.vi.1981, E. Tuzlacı (ISTE 46494)! Osmaniye: Sorkun yaylası, Amanos da., 400 m, 21.vı.1967, Y. Akman 7721 (ANK)! C7 Adıyaman: Kahta-Sincik, Cendere köprüsünün 1 km ilerisi, Kahta'ya 22 km, 680 m, killi tepeler, 15.vi.1979, E. Tuzlacı, M. Saraçoğlu (ISTE 42247)! C8 Mardin: Mardin-Deyrulzafaran kilisesi, kilise yakınındaki kıraç yamaçlar, 880 m, 9.vı.1979, E. Tuzlacı, M. Saraçoğlu (ISTE 42127)! C9 Siirt: Eruh-Şırnak, Eruh'a 12 km, Yanılmazı kö. yakınları, 1200 m, 18.v11.1981, E. Tuzlacı (ISTE 47387)!



Adsorption of iron, lead, paracetamol, imipramine on natural polymers

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Abstract

Poisoning results from many reasons such as misuse or overdose of drugs. Suicidal and murder purposes are mostly severe, serious and life-threatening cases which require immediate intervention and treatment. Among lifesaving methods external and/or internal decontamination is the most important. Internal decontamination (gastrointestinal) is an effective process for intoxication control that can be done by adsorptive materials. Activated charcoal is used as unique local antidote for adsorption of causative agent. Considering their significance and effectiveness, adsorptive materials are necessary to be developed. In the present study, starch and naturally extracted pectin from citrus, in the presence of trace amount of potassium per sulphate as initiator, were thermally grafted to chitosan to form natural, inert, and highly adsorptive polymeric surfaces. This polymer is convenient for biomedical purposes. Upon drying at 37°C for 48 hours, thermally cross-linked products were obtained. FTIR, UV-Visible spectrophotometer and SEM analyses were applied in order to characterize the products. To evaluate the adsorption potency of new adsorptive material, lead and iron which cause common poisoning were applied on the polymers. The results showed that adsorption degree of lead and iron were maximum 50% and 30% respectively. Desorption amounts can be a sign of adsorption potency. In this study, paracetamol and imipramine, which are commonly used drugs that can and caused intoxication in case they are misused or use for suicidal purpose were applied onto two polymers which contain pectin desorption amount for two drugs were determined. SEM pictures taken before and after blood/polymer contact didn't reveal any significant blood component attachment on the chitosan-graft- (starch; pectin) film surface. Indicating no hemocompatibility.

Keywords

Chitosan, decontamination, hemocompatibility, imipramine, iron, lead, paracetamol, pectin, starch.

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Intentional and unintentional exposure to various types of drug, chemicals, residuals, wasting materials, herbal products, food contaminants and environmental pollutants at high doses via many ways are common reasons for health problems especially in human (Crowl and Louvar 2001). As a result, many different type poisonings can be seen. Intoxication approaches that consist of airway, breathing, circulation and decontamination (ABCD) are lifesaving and important.

Decontamination has both external and internal types. Internal decontamination of gastrointestinal system is very important in poisoning. For this purpose, some processes such as stomach washing and emesis are applied. Furthermore, adsorption of causative substance would be useful and effective.

Many drugs, chemicals and causative agents are easily adsorbed to local antidote at considerable amount. On the other hand, some others don't show affinity to local antidote. One of the best-known adsorptive materials is activated charcoal that has a wide range of adsorptive potency to different materials and drugs. Development of such adsorptive material with higher affinity will be very useful.

For this reason, in the present study some polymers were prepared as new adsorptive materials. Heavy metals like lead (Pb), essential metal iron (Fe), paracetamol as analgesics and a tricyclic antidepressant, imipramine, were analyzed to evaluate their adsorptive degree to these new polymers.

Although some metals are essential for the body heavy metals which cause toxic effects in human can be very harmful (Geiger and Cooper 2010). Lead exposure happens through air and food in almost the same amounts. More than 50% of lead productions are produced from petrol. Industrial lead exposure happens in mines and smelters, together with lead painted metal repairing, and in battery plants. Glass industries are responsible for low or moderate exposure. High levels of air discharges may lead to pollution in areas near lead mines and smelters. Soil and water can be contaminated by airborne lead deposition which may lead to human exposure via food chain.

Lungs absorb up to 50% of inhaled inorganic lead. Adults absorb 10–15% of the lead inside food, while children's gastrointestinal tract may absorb up to 50%. Lead is bound to erythrocytes in blood, and has a slow elimination rate via urine. Lead accumulates in the skeletal system and has a slow releasing rate from different tissues and structures. The half-life of lead is about 20– 30 years and 1 month in skeleton and blood, respectively. Organic lead compounds can enter the body and cell membranes (Järup 2003). Tetramethyl and tetraethyl lead have good skin penetration. These compounds also have the ability to cross the blood-brainbarrier in adults. That's why organic lead compounds may cause acute poisoning and encephalopathy in adults in consequence. Blood-brain-barrier permits the entrance of organic lead in adults. In babies the crossing ratio to brain is higher than adults. The high gastrointestinal uptake and the penetrable blood-brain-barrier make children good candidates for brain damage related with lead exposure. Headache, irritability, abdominal pain, proximal renal tubular damage and other different nervous system related symptoms are the symptoms of acute lead poisoning. Sleeplessness and restlessness the important are most of lead encephalopathy. symptoms Behavioral disturbances and learning and concentration difficulties are the results of lead poisoning in children (Järup 2003). Patients with lead encephalopathy may experience acute psychosis, confusion and reduction in consciousness. People who have been under lead exposure for a long time undergo memory deterioration, may prolonged reaction time and reduction in understanding ability. Individuals whose blood lead levels are under 3 µmol/l may express peripheral nervous symptoms with decreased nerve transference velocity and dermal responsiveness. If severe neuropathy occurs, the lesion may stay for life long. In less severe circumstances. the most

noticeable sign of lead poisoning is hemoglobin synthesis disturbance, and longterm lead exposure may lead to anemia and kidney damage (Järup 2003). Inorganic lead shows toxicity in nephrons because of its high affinity to accumulate in proximal tubule. Kidney damage can occur as result of long-term lead exposure. According to a recent study including Egyptian policemen, NAG excretion was directly shown to be related with the duration of lead exposure (Gang *et al.* 1988).

Iron in trace amounts is essential element for body. Iron overdose may cause serious poisoning in which symptoms usually appear within 6 hours after poisoning. Iron poisoning occur at 5 stages which may result in life threating circumstances such as destructive damages to gastrointestinal (GI) mucosa, hemorrhagic gastritis, considerable fluid loss, bleeding and shock (Baranwal and Singhi 2003).

Moreover, drugs which are used at normal therapeutic doses for diagnosis, therapy and poisoning treatment, can also lead to poisoning (Prescon 1983).

Normally paracetamol (acetaminophen) is known as a safe drug but since it's an OTC drug which is consumed in large amount by adults and children, its toxicity which may cause acute centrilobular hepatic necrosis is common. Paracetamol poisoning has no specific or early signs and symptoms and doesn't lead to impaired consciousness. Acute renal failure which is an uncommon complication may also occur (Prescon 1983; Haddah *et al.* 1983).

Imipramine is an antidepressant medication that is a cheap and accessible. Therefore, its accidental consumption or use for suicidal purposes is common. The toxicity signs include, slow breath, low blood pressure, rapid heartbeats and disturbance of electrocardiograph (Brush and Aaron 2007). An adsorptive blood compatible polymer must be inert, non-reactive, form's strong bonds and possess high affinity to agents. The polymer-agent complex should be easily excreted by feces (Bahramzadeh *et al.* 2019).

MATERIALS AND METHODS

Following materials, instruments and methods were used to prepare and characterize the polymers: Chitosan medium molecular weight (450 kDa) with degree of deacetylation of 85% (Sigma-Aldrich), corn starch, lemon extracted from citrus fruit potassium per sulfate (KPS) (EDH Chemicals LTD), (Titrachem), acetic acid (Sigma-Aldrich), acetone (Tekkim Kimya San), ethanol (Selim ve Oglu Ltd), hydrochloric acid (Merck) without any purification, lead two oxide (Sigma-Aldrich), iron three chloride (Sigma-Aldrich). dithizone (Sigma-Aldrich), salicylic acid (Sigma-Aldrich), paracetamol tablet (Minoset), imipramine tablet (Novartis 25 mg) and methanol (Sigma-Aldrich). The products were characterized by FTIR (Perkin Elmer, Spectrum Two Spectrometer), UVvisible spectrophotometer (Perkin Elmer) and scanning electron microscope (SEM) VP (LEO 1450 Scanning Electron Microscope). SEM analysis was carried out in Ferdowsi University, Mashhad, Iran.

Preparation of new adsorptive material Pectin extraction

25 g of lemon was weighted and transferred in a 250 ml beaker. 100 ml of water and 1.5 ml of 3 N sulphuric acid (H₂SO₄) solution were added. The mixture was heated by magnetic stirrer. Temperature of the mixture was controlled by a thermometer and heated at 85-90 °C for 30 minutes. Following heating, the aqueous part was filtered through a cotton swab of cotton wool. Cotton was squeezed with the help of baguette. The remaining crusts were further extracted for 15 minutes under the same conditions. The filtrates were combined. The filtrate was transferred to a 500 ml sieve and 96° ethanol was added until the alcohol grade became 55°. The precipitated pectin was filtered under Buchner funnel under a slight vacuum. The precipitate was completely transferred to a funnel and washed firstly with 25 mL of 96° alcohol and then continued washing, for two times with 15 mL of acetone.

Preparation of films

Specific amount of starch and pectin were dissolved in 15 mL of chitosan solution (1% w/v solution in 1% v/v acetic acid solution) at room temperature, as shown in Table 1. The mixture was transferred to a petri dish, and a film layer was formed due to thermal crosslinking, following the evaporation of solvent at 37°C for 48 hours. Dried samples were taken and impurities were cleaned off the films by immersing in water. Grafted products were named as chitosan-*graft*-(starch; pectin).

Under similar experimental conditions, in the absence of pectin, a mixture of chitosan/starch solution was also prepared in a form of film.

Table 1: Synthesis of chitosan-graft-starch and chitosan-graft-(starch; pectin) films*.

Film	Pectin (ml)	Starch (g)
F1	0	0.2
F2	2	0.2
F3	5	0.2

*15 mL of 1% v/v acetic solution, 0.15 g chitosan, 0.075 g KPS, 37°C, 48 hours.

Swelling kinetics

From each of dry thermally cross-linked sample film samples, 0.01 g was taken, soaked in water. The weight was recorded in every 30 minutes (leaking was avoided). Swelling percentage was calculated with respect to the following equation (Bahramzadeh *et al.* 2019):

Swelling % = $\frac{Ws - Wd}{Wd} \times 100$; eq. (1)

where Ws (g) and Wd (g) stands for the weights of swelled and dry hydrogels, respectively.

In-vitro platelet adhesion analyses

Films were covered by human fresh blood obtained from healthy donors, washed by ultra-pure water and dried to examine contact properties by SEM (Caner *et al.* 1998).

Lead adsorption by chitosan-*graft*-(starch; pectin) films

Films (0.05 g) were covered by 10 ml Pb²⁺ solution at 125, 250 and 500 ppm concentrations at 1, 2, 3 and 24 hour time intervals. 2 ml Pb²⁺ solution from each test tube was taken and mixed with 1 ml alcohol dithizone solution. Resultant absorbance values was recorded, at room temperature at 472 nm wavelength. Triplicated measurement was applied for each sample and the average value was recorded. Percent removal of Pb²⁺ was calculated according to following equation:

$$\frac{(Ai - Af)}{Ai} \times 100$$
; eq. (2)

where Ai is initial absorbance and Af shows final absorbance.

Fe³⁺ adsorption by chitosan-*graft*- (starch; pectin) films

0.01 g of the film was added to 4 different test tubes and covered by 4 ml Fe³⁺ solution at the concentrations of 125, 250, 500, 900 ppm and incubated for 1, 2, 3, 4 hours. From each of the test tubes, 0.5 ml Fe³⁺ solution was drown and mixed with 0.5 ml of 5sulfosalicylic acid dehydrate, (10% w/v), and the volume was completed upto 4 ml using pH = 1 buffer solution. Absorbance was recorded by visible spectrophotometry at 505 nm. Finally, percent removal of Fe³⁺ was calculated according to equation 2.

Paracetamol loading into chitosan-graft-(starch; pectin) films

0.01 g of the film was added to 4 different test tubes and covered by 4 ml paracetamol

solution at the concentrations of 3.75, 7.5 and 15 ppm and remained for 48 hours. After 48 hours, the films were taken from solution and dried. Afterwards, they were placed in a test tube and 4 ml distilled water was added to release the adsorbed paracetamol. Desorption results were monitored by UV-Visible spectrometer at 242 nm in 5 hours with 1 hour intervals.

Imipramine loading into chitosan-*graft*-(starch; pectin) films

0.01g of the film was added to 4 different test tubes and covered by 4 ml imipramine dissolved in methanol at the concentrations of 0.077, 0.155, 0.31 ppm. After 48 hours, the films were taken and placed in a test tube covered with 4 ml methanol and the desorption results were taken by visible spectrometer at 251 nm.

RESULTS

Figure 1 shows photograph of Chitosan-graft- (starch; Pectin) films.



Figure 1: Chitosan-graft- (starch; Pectin) films.

Table 2 shows at (37°C) the percent of grafting increases while the amount of pectin increased.

Sample ID	Starch (g)	Pectin (g)	Grafting % 37°C
S1	0.2	0	0
S2	0.2	2	23
S3	0.2	5	43

 Table 2: Synthesis of Chitosan-graft- (starch; pectin) Films.

Swelling behavior of chitosan-graft-polyHEAA and chitosan-graft-(polyHEAA;MBA) films

Figure 2A and 2B show the swelling behavior of chitosan-*graft*- (starch; pectin) films with different amount of pectin. They both showed maximum 70% swelling but at different time intervals. S3, standing for a compound with higher amount of pectin, marked maximum swelling after 30 minutes whereas S2, representing the same

compound containing less amount of pectin showed maximum swelling after 90 minutes. They both increased the biodegradability time because samples lasted for a longer period. This promises the natural modification for drug loading and adsorbing toxins. On the other hand, S1 was degraded far faster and less controllable.

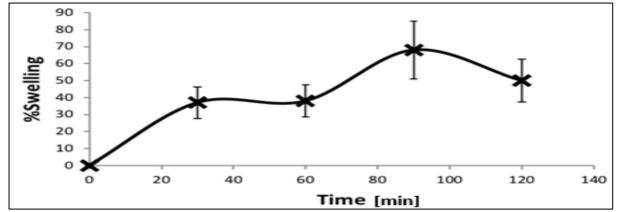


Figure 2A: Swelling behavior of S2 (Chitosan-graft-(starch; 2g Pectin) films at pH=7.4 in 2 hours within 30 min interval.

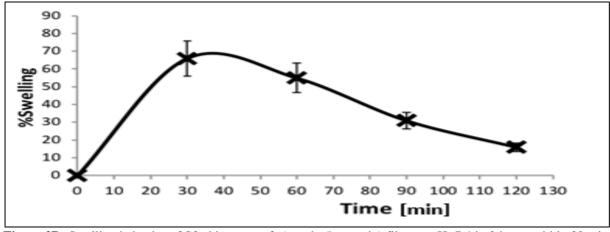


Figure 2B: Swelling behavior of S3 chitosan-graft-(starch; 5 g pectin) films at pH=7.4 in 2 hours within 30 min interval.

FT-IR Analysis

Films characterized by **FTIR** were spectrometer to assess modifications. Figure 3a shows major functional groups for chitosan where broad band after 3000 cm⁻¹ represents H-bonding and 2 picks on 1551 and 1649 cm⁻¹ stands for C-O and amide functional groups, respectively. Moreover, 2 picks at 2884 and 2974 cm⁻¹ show C-H stretching. However, a pick at about 1370 cm⁻¹ appeared when the polysaccharide chains were grafted onto the chitosan backbone which were concluded as C-H bond. In the FTIR spectrum of chitosanstarch shown in Figure 3b, amide band at 1649 cm⁻¹, C–H bending vibrations in the 1400–1500 cm⁻¹ region, $-CH_3$ bending at 1380 cm⁻¹, C-H stretching at 2884 and 2974 cm⁻¹ and O-H stretching at 3339 cm⁻¹ were recorded.

When it comes to the pectin containing films, (Figure 3c and 3d), all the previous spectra were similar except between 1630 cm⁻¹and 1747cm⁻¹. Moreover, an additional signal appeared which became more intense once the pectin amount had been increasing (Figure 3d).

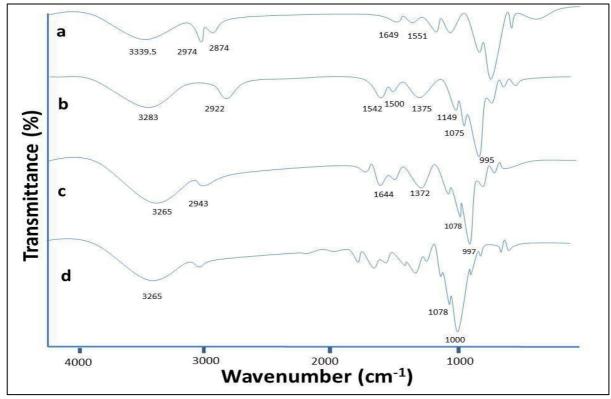


Figure 3: The FTIR spectrum of a. chitosan b. S1 (chitosan grafted starch) c. S2 (chitosan-graft-(starch; 2 g Pectin) films d. S3 (chitosan-graft-(starch; 5 g pectin) films.

SEM analysis for chitosan-graft-(starch; pectin)

The surface morphologies of grafted products were examined using SEM pictures as shown in Figure 4A, 4B and 4C. SEM images of samples exhibit more spherical structures on the surface and therefore wider surface area and more adsorption potency due to excessive exposed active sites, in the presence of pectin in samples S2 and S3. On the other hand, less fine spherical structures were detected in S1 where starch was grafted on to chitosan. In brief, addition of pectin showed a positive modification role when its entire surface area was taken into account.

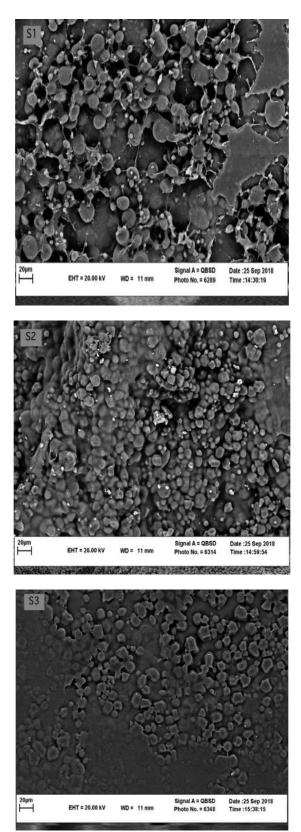
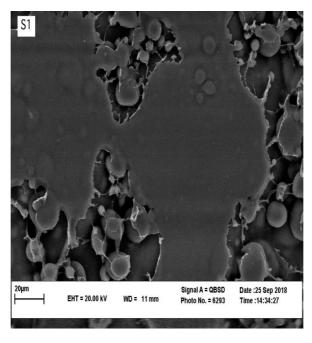


Figure 4A: SEM pictures (200 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (2 g) (S2), and starch (0.2 g) chitosan-graft-pectin (5 g) (S3),



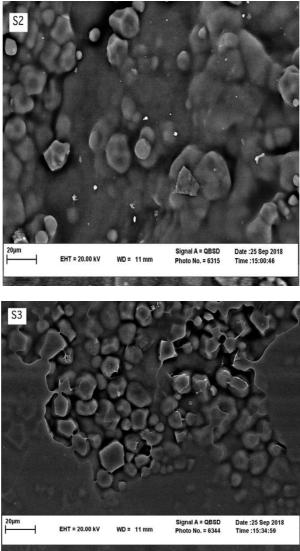


Figure 4B: SEM pictures (2000 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (2 g) (S2), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

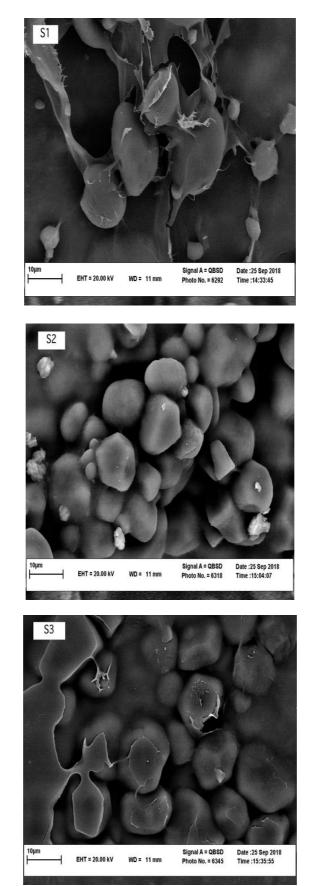


Figure 4C: SEM pictures (5000 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (2 g) (S2), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

Figure 5 shows the SEM images for films when they come in contact with blood in vitro conditions. The surface of polymers does not exhibit any notable different texture. Any sign of blood coagulation is not detectable before and after blood contact. However; denser matrix, as a result of physical blood stream, seems to be loaded in pores of films.

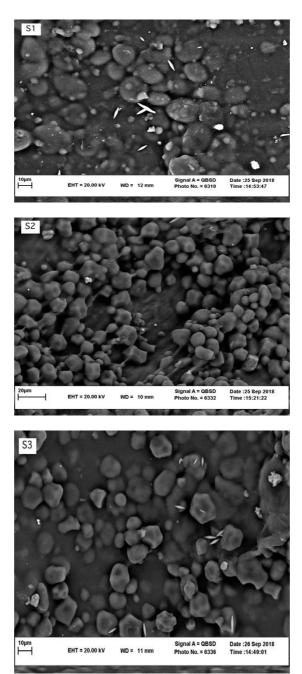


Figure 5: SEM picture after blood contact (2000 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (2 g) (S2), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

Lead adsorption on chitosan-*graft*- (starch; pectin) films spectrometric adsorption analysis of lead on chitosan-*graft*- (starch; pectin) films

Tables 3A, 3B and 3C show lead removal percentage from lead solutions at various concentrations.

The results show that as the amount of pectin increases, higher amounts of lead were adsorbed by the films. This evidence was generally and specifically confirmed by SEM. The SEM pictures are exhibited in Figure 4A, 4B and 4C. Despite some fluctuations at low concentration, S3 was **Table 3A:** Pb²⁺ removal at 125 ppm solution. adsorbed well whereas at high concentration S2 was dominating adsorbent. In order to monitor lead adsorbtion, SEM pictures were taken. As a result, lead was adsorbed more on the surface of chitosan-*graft*-starch films when compared to chitosan-*graft*-(starch; pectin) sample that showed intensive adsorption. This could be the result of higher lead adsorption potency of pectin containing films.

	Removal % (1	125 ppm)	
Time (h)	S1	S2	S3
0	0	0	0
1	0	1,531729	1,531729
2	32,82276	43,32604	55,57987
3	21,00656	27,78993	41,57549
24	17,72429	23,63239	38,0744

Table 3B: Pb removal at 250 ppm solution.

Removal % (250 ppm)			
Time (h)	S1	S2	S3
0	0	0	0
1	2,534113	0,584795	2,534113
2	23,97661	20,07797	22,02729
3	37,4269	27,87524	38,79142
24	-6,23782	26,90058	13,84016

Table 3C: Pb removal at 500 ppm solution.

Removal % (500 ppm)			
Time (h)	S1	S2	S 3
0	0	0	0
1	7,749077	9,409594	8,671587
2	9,225092	10,88561	8,671587
3	11,43911	12,36162	9,594096
24	28,78229	31,18081	27,12177

Figure 6A, 6B and 6C exhibit lead adsorption on low pectin and high pectin polymers in different magnifications.

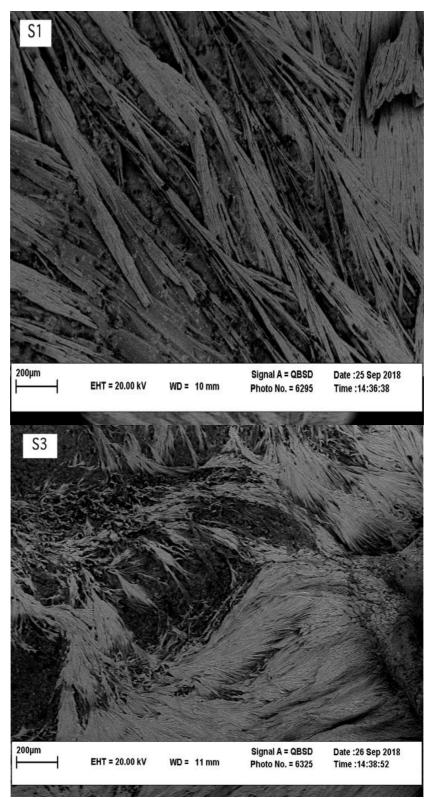


Figure 6A: SEM pictures of lead adsorption (200 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

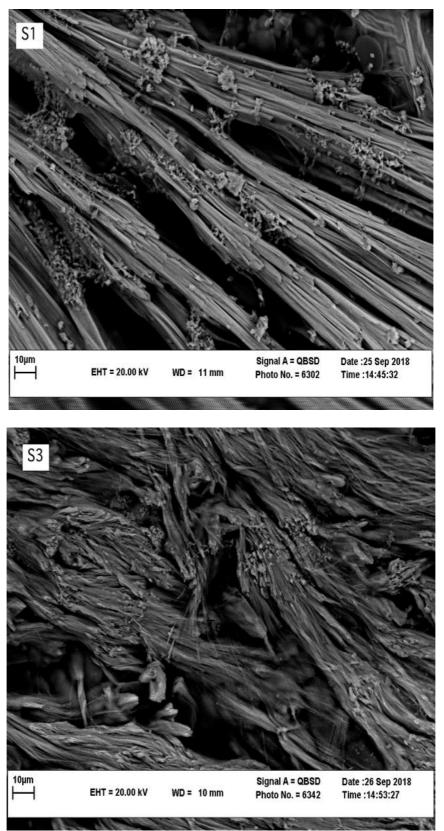


Figure 6B: SEM pictures of lead adsorption (2000 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

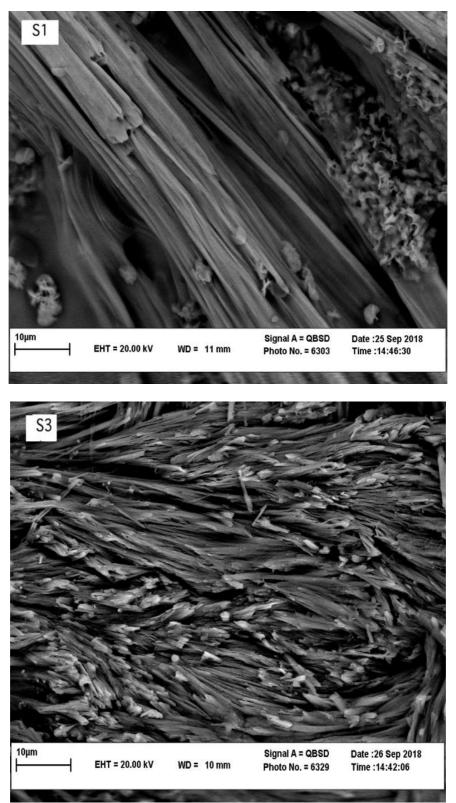


Figure 6C: SEM pictures of lead adsorption (5000 times magnified) of starch (0.2 g) chitosan (S1), starch (0.2 g) chitosan-graft-pectin (5 g) (S3).

Iron (Fe³⁺) adsorption

Table 4A, 4B, 4C and 4D show iron adsorption at different concentrations. Except 125 ppm which low amounts of pectin showed higher adsorption potency in comparison with S1 (no pectin), at other concentrations all samples show some degree of adsorption but generally S1 was the best adsorbent.

Removal % (125 ppm)			
Time (h)	S1	S2	S3
0	0	0	0
1	9,558824	25	15,44118
2	-1,47059	37,5	13,97059
3	8,088235	18,38235	15,44118
4	14,70588	25,73529	18,38235

Table 4A: Fe removal at 125 ppm solution.

Table 4B: Fe removal at 250 ppm solution.

Removal % (250 ppm)			
Time (h)	S1	S2	S 3
0	0	0	0
1	28,48665	32,64095	16,32047
2	29,67359	34,7181	28,18991
3	16,32047	13,05638	6,824926
4	33,53116	23,1454	16,91395

Table 4C: Fe removal at 500 ppm solution.

Removal % (500 ppm)			
Time (h)	S1	S2	S3
0	0	0	0
1	4,761905	11,72161	10,80586
2	22,71062	14,46886	16,66667
3	12,08791	13,00366	10,25641
4	29,30403	31,68498	27,65568

Table 4D: Fe removal at 900 ppm solution.

Removal% (900 ppm)			
Time (h)	S 1	S2	S3
0	0	0	0
1	4,761905	11,72161	10,80586
2	22,71062	14,46886	16,66667
3	12,08791	13,00366	10,25641
4	29,30403	31,68498	27,65568

Paracetamol

Table 5A, 5B and 5C show drug loading results after 2 hours of contact time. chitosan-graft- (pectin; starch) films released more paracetamol than chitosan-graft-starch films in certain time intervals which means pectin containing films have higher desorption potency and indirectly can be a sign of higher adsorption potency.

Table 5A: Desorption of sample placed at 3.75 ppm paracetamol solution.

Time (h)	S1	S2	S3
1	0.124	0.105	0.124
2	0.147	0.155	0.168
3	0.156	0.179	0.198
4	0.185	0.178	0.202
5	0.182	0.190	0.219

*Initial: 0.233

 Table 5B: Desorption of sample placed at 7.5 ppm paracetamol solution.

Time (h)	S1	S2	S 3
1	0.138	0.107	0.116
2	0.154	0.155	0.163
3	0.175	0.182	0.190
4	0.183	0.180	0.230
5	0.186	0.187	0.211

*Initial: 0.464

Table 5C: Desorption	of sample placed at 15	ppm paracetamol solution.
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Time (h)	S1	S2	S 3
1	0.127	0.109	0.108
2	0.156	0.158	0.157
3	0.178	0.187	0.186
4	0.174	0.192	0.238
5	0.190	0.225	0.224

*Initial: 0.233

Imipramine

Table 6A, 6B and 6C shows drug loading results after 2 hours of contact time. Generally, imipramine loaded films follow the same patterns of paracetamol loaded films. This shows that pectin containing films have higher desorption potency for imipramine and indirectly can be a sign of higher adsorption potency.

Time (h)	S1	S2	S 3
1	0.188	0.153	0.152
2	0.189	0.169	0.198
3	0.197	0.155	0.206
4	0.206	0.169	0.225
5	0.217	0.183	0.207

Table 6A: Desorption of sample placed at 0.31 ppm imipramine solution.

*Initial: 0.956

Table 6B: Desorption of sample placed at 0.155 ppm imipramine solution.

Time (h)	S1	S2	S 3
1	0.164	0.162	0.166
2	0.157	0.193	0.156
3	0.155	0.181	0.186
4	0.176	0.184	0.197
5	0.190	0.212	0.203

*Initial: 0.538

Table 6C: Desorption of sample placed at 0.077 ppm imipramine solution.

Time (h)	S1	S2	S3
1	0.146	0.149	0.164
2	0.171	0.180	0.168
3	0.155	0.181	0.186
4	0.165	0.175	0.179
5	0.166	0.197	0.197

*Initial: 0.357

DISCUSSION

Poisoning cases are very common due to improper drug usage all around the world. These types of cases require immediate curative application and treatment. Among these, decontaminations is very important. For this purpose, adsorbant material like activated charcoal as a local antidote is used. In previous studies, the polymer without pectin was prepared (Hasipoglu *et al.* 2005; Caner *et al.* 2007; Yilmaz *et al.* 2007; Adali and Yilmaz 2009; Yilmaz *et al.* 2016). However, in the present study almost the same polymer with high and low amount of pectin were assayed. The present polymer (thermally grafted natural chitosan-(starch; promising pectin) showed adsorptive potency with respect to natural adsorbent advantages for Pb⁺², Fe³⁺ removal and also for paracetamol, imipramine drug loading properties. Pectin grafted polymeric film shows up to 50% Pb^{+2} and up to 34% Fe^{3+} adsorption potency. It also shows up to 46% and 36% better releasing property for paracetamol and imipramine, respectively, after 48 hours of drug loading which indirectly can be a sign of better absorbance capacity.

According to the results of present study, the grafted polymer may be a good candidate as a local antidote for internal decontamination in the treatment of drug and metal poisoning due to it's natural, blood compatible, costeffective, high chelating and sustained released nature.

Internal decontamination is a very important and effective process for intoxication control that can be done by adsorptive materials.

It should be non-thrombogenic and nonhemolytic which can be found in pectin and starch. Since chitosan is known for its biocompatibility, pectin for its being adoptive, environmentally friendly and having controlled release and starch for its having hydrogen accepting and polymeric properties, the material has got the potential to improve the surface and bulk properties of chitosan as a biomaterial. As synthesis of chitosan-graft starch pectin copolymers have not been reported in the literature before, this study aimed to find out if natural thermal grafted polymers have the adsorptive and drug loading properties. If so, to identify the optimum process conditions, and to characterize physicochemical the characteristics of the products. Although the adsorbent properties of pectin had been studied before, it has not been investigated in terms of it hemocompatibility as an adsorptive matrix grafted to chitosan and starch. The polymer may be applicable for soil, water, air decontamination purposes and worths the further studies.

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Analgesic nephropathy

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Abstract

Non-steroidal anti-inflammatory drugs (NSAIDs), which are easily accessible, inexpensive and have many therapeutic effects, are used frequently and in large quantities by every age group of patients. NSAIDs reversibly inhibit cyclooxygenase enzymes at various degrees and they have the same action and toxicity mechanism. These groups of drugs can cause damage in many organs when there used at high dose and for a long-time period. Analgesic nephropathy is one of the prominent side effects of NSAIDs which damage kidneys. The present review focused on renal toxic mechanisms induced by NSAIDs. The geriatric group which is the most vulnerable group to misusage of NSAIDs should be well informed and monitored by healthcare professionals to decrease the risk of adverse effects related to NSAIDs.

Keywords

Analgesic, analgesic nephropathy, cortical tubulointerstitial nephritis, non-steroidal anti-inflammatory drugs, renal papillary necrosis.

Article History

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INTRODUCTION

Analgesic nephropathy is a kind of chronic renal incompetency which is caused by long term usage of one or more analgesic for medication. The specific medication and the dose interval required have not been fully understood (De Broe 1998). Analgesic nephropathy can be considered as one of the causes of end-stage renal failure (De Broe 1998; Chang *et al.* 2008). These patients generally had been taken analgesics for months or years because of chronic pain like headaches or backaches (Gault and Wilson 1978). Its pathological signs; include atrophic kidneys, renal papillary calcification, and irregular renal contour.

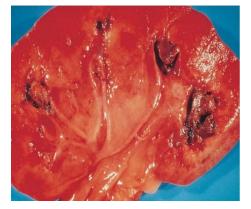




Figure 1: Pathological signs of analgesic nephropathy (Pinter et al. 2004; Noels et al. 1995; De Broe 2009).

A relation between non-steroidal antiinflammatory agents (NSAIDs) and chronic kidney disease has long been under investigation. NSAIDs are the most commonly used medicines in the treatment of pains, inflammations and fever. NSAIDs are accessible, cheap, and can be sold with or without prescription (Buer 2014).

Chronic pain is a treatable condition that at any one point in time affects 20%–46% of community-dwelling older adults and 28%– 73% of residents in aged-care facilities. A number of practice guidelines and literature reviews related with to the management of chronic pain in the elderly patients suggest paracetamol as the first-line management option (Abdulla *et al.* 2013). Several casecontrol studies have reported associations between chronic renal failure and other analgesic preparations, including aspirin, antipyretics, and NSAIDs in combination with caffeine, codeine, and/or barbiturates (De Broe 2009). It has been proposed that paracetamol, not phenacetin, accumulated in the renal papilla, and in animal experiments phenacetin appeared to be less nephrotoxic than the other analgesics (Bluemle 1969).

In the 1950s, researchers in Sweden and Switzerland illustrated that renal papillary necrosis developed from long term ingestion of large number of phenacetins containing analgesic mixtures. During the 1960s and 1970s, phenacetin was singled out as the nephrotoxic agent in the analgesic mixtures, leading to its ban in several countries. Thereafter, the incidence has declined markedly. The nephrotoxicity of phenacetin is dose dependent. An intake of 6–8 tablets per day for a period of 6–8 years lead to the development of AN (Nanra *et al.* 1987).

Continuous and persistent consumption of NSAIDs cause many adverse effects in the body. Especially, prolonged use can cause damage to the kidneys and lead to the development of nephropathy. The major aim of present study was to review nephrotoxicity mechanisms induced by NSAIDs. For this purpose, many literatures were reviewed and under the light of these literatures the mechanisms of analgesic nephropathy were evaluated in detail.

Therapeutic effect mechanism of NSAIDs NSAIDs reversibly inhibit cyclooxygenase (COX) enzymes at various degrees. Prostaglandins are responsible in functional regulation of several organs such as the gastrointestinal tract (maintaining GI mucosal integrity), increase renal blood flow, promote blood clotting by activating platelets, and also affect kidney function (Hawkey 2001). However, excessive number of prostaglandins are responsible for enhancing fever, inflammation and pain. Therefore. inhibiting prostaglandin production can cause adverse effects even in the therapeutic range of NSAIDs usage. There are two types of isoenzymes, COX-1 and COX-2, which were identified in early 1990s and known to be involved in production of prostaglandin. The main role of COX-1 is to produce prostaglandins and COX-2 become induced in response to inflammation (Hilário et al. 2006). Figure 2 shows the arachidonic acid pathway. The location of isomerases of the prostaglandins have be shown in Figure 3.

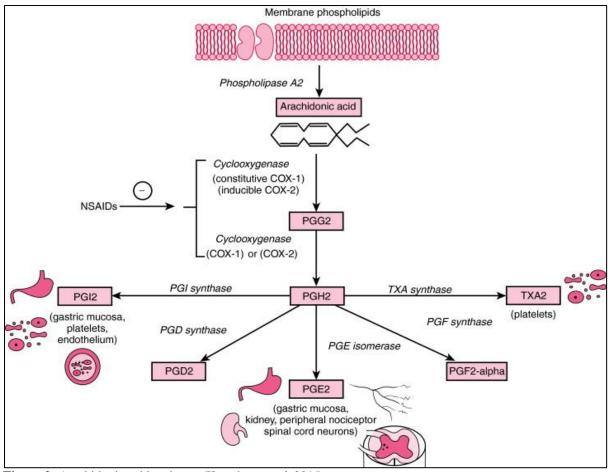


Figure 2: Arachidonic acid pathway (Kawahara et al. 2015)

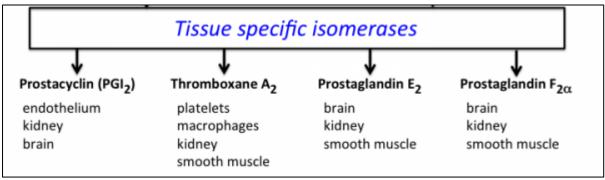


Figure 3: Prostaglandin isomerases specific to each tissue (Kawahara et al. 2015).

Many of the NSAIDs act nonspecifically on COX but more recently developed NSAIDs have been reported to act more specifically on the COX-2 isoenzyme, with the aim to decrease fever, pain and inflammatory response whereas reducing associated gastrointestinal and renal side-effects relating to COX-1 inhibition. According to recent studies, COX-2 selective agents such as Rofecoxib and Celecoxib can promote thrombosis and substantially increase the risk of heart attack (Serhan and Levy 2003). However, the pattern of toxicity is the same with COX-2 selective and COX nonselective (e.g. aspirin) NSAIDs at overdose cases. These medicines act by binding to the active sites of COX and preventing the catalysis of arachidonic acid (AA) to prostaglandins therefore exerting analgesic, antipyretic, and anti-inflammatory properties (Hunter et al. 2011). The chronic use of NSAIDs at therapeutic doses is generally safe in patients with normal physiology and without any underlying problem. Toxicity associated with NSAIDs is result of excessive inhibition of COX-1 and eventually reduction in prostaglandin synthesis. Therefore, this condition can trigger such organ damage as gastrointestinal, renal and central nervous systems (CNS) as adverse effects both in therapeutic use and in acute overdose (Hunter *et al.* 2011).

Nephrotoxicity mechanisms of NSAIDs

NSAID-induced renal adverse effects are rare, sometimes temporary and usually reversible at the time of drug withdrawal. The occurrence rate and the seriousness of the renal effects raise in patients with risk determinants such as those who has diabetes, heart failure, cirrhosis, renal dysfunction, users of diuretics and in the elderly subjects. Unwanted effects extend from electrolyte retention and reduce glomerular filtration (by means of inhibiting vasodilator prostaglandins) to nephritic syndrome and chronic renal failure. While acetaminophen and aspirin may develop chronic interstitial nephritis, remaining NSAIDs may generate acute interstitial nephritis, altered intraglomerular hemodynamics, chronic nephritis interstitial and glomerulonephritis. The correlation has found between been high plasma concentration and the renal adverse effect of NSAIDs (Emkey et al. 1982). Early diagnosis is important because chronic interstitial nephritis has been known to progress to end-stage renal disease (Harirforoosh and Jamali 2009).

In cases of severe toxicity, detoxification and compensatory mechanisms are insufficient and an increase in rate of kidney insufficiency can be encountered. It has been reported that over 300 chemical substances including NSAIDs cause kidney damage and the incidence of drug-induced nephrotoxicity is as high as 66 % in elderly patients (Murray *et al.* 1971).

Nephropathy can be the result of oxidative stress, impaired membrane integrity, direct toxicity because of chromium or cadmium accumulation. While the different parts of nephron may be the targets of nephrotoxins, the proximal tubule is the most common target site of nephrotoxins, including the drugs (Murray *et al.* 1971).

Glomerular filtration is reduced because of the increase in the amount of tubular cell death. If this happens too many nephrons, the total glomerular filtration rate would decrease and tubular cell loss leads to the abrasion of the basolateral membrane which would prevent clearance of compounds that are required to be removed from the body by urine. This may lead to acute renal failure within a few days. Severe DNA damage can also cause cell death. If DNA damage that is not very severe is repaired improperly, the remaining DNA lesions may lead to the formation of renal cancer over the years. Long-term treatment with opioid analgesics may not cause direct damage to the kidney. However, it may induce the infiltration of immunocytes to kidneys over a long time. While these immunocytes do not directly produce acute nephritis, the frequent use of these agents may gradually reduce renal function (Schrier 2013).

On the other hand, kidney has a great regeneration ability. Even if a large part of the proximal tubular cells is lost, the cells multiply within 1-3 days after the initial injury. The new cells are flat and within a few days they will differentiate into unique proximal tubule cell (Schrier 2013).

NSAIDs cause nephrotoxicity that can present as various renal syndromes such as acute kidney injury, nephritic syndrome, interstitial nephritis, and chronic renal failure (Figure 4). The common link among these various syndromes is the disruption of PG synthesis. The PGI2 has predominant vascular actions in the form of renal vasodilation. PGE2 has renal tubular action in the form of inhibition of salt and water reabsorption, especially in the thick ascending limbs of loop of Henle and collecting ducts (Schrier 2013).

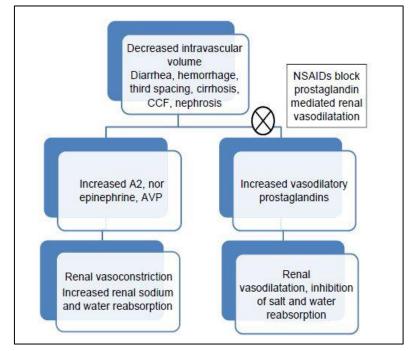


Figure 4: Mechanism of renal toxicity of NSAIDs (Schrier 2013).

These biological actions of the autacoids on the kidney are prominent in the setting of a decreased effective arterial blood volume during which there is an increase in the circulating levels of angiotensin II (AII), arginine vasopressin (AVP) and catecholamines. PGs, once released, act to counterbalance the effect of the abovementioned hormones by causing renal vasodilation and inhibition of salt and water reabsorption. Therefore, the inhibition of PG synthesis by NSAIDs (COX inhibitors) leads to the unopposed action of AII, AVP, and catecholamines, resulting in enhanced renal vasoconstriction and salt and water reabsorption. As the renal medulla is dependent on the production of PGs for its blood flow, the inhibition of PG synthesis by NSAIDs leads to medullary ischemia and papillary necrosis (Schrier 2013).

In order to comprehend the renal effects of NSAIDs, a brief knowledge of the physiologic function of eicosanoids in the kidney is essential. A composite term, eicosanoid, is the outcome of arachidonate other polyunsaturated fatty or acid metabolism and covers prostaglandins, prostacyclin, thromboxane, epoxides and lipoxygenase products. COX isoforms are responsible for this conversion and they are inhibited irreversibly by aspirin and reversibly by the remaining NSAIDs (Schrier 2013). There are four main areas within the kidney that are rich in the

afferent and efferent arterioles, interstitial cells of renal medulla, the glomerulus and collecting medullary ducts. Eventual products of the conversion are further metabolized to PGI₂, PGE_{2alpha}, PGD₂ and TXA₂. This suggests that each cell which has cyclooxygenase enzyme can create all types of eicosanoids, however one or two are predominant depending on the situation. Accordingly, arterioles produce PGI₂, glomeruli produce PGI₂, PGE₂ and TXA₂, and both interstitial cells and collecting duct cells produce PGE2. Both TXA₂ and PGH₂ have practically same actions and they activate a mutual receptor (Schrier 2013). Their binding ends up with phospholipase C stimulation, creation of inositol diacylglycerol and triphosphate, and super elevation of free cytosolic calcium derivatives from intra- and extracellular origins. Subsequently, smooth-muscle contraction and renin release inhibition are generated. In contrast, PGI₂, causes vasodilation and renin release stimulation by activating adenyl cyclase and raising intracellular adenosine cyclic monophosphate (Lote et al. 1989).

concentrations of both isoforms of COX; the

Phospholipase A₂ which is found in renal arterioles and is activated by circulating vasoconstrictors like angiotensin II and norepinephrine, leads to production of PGI₂, that adjusts renal vasoconstriction. The lipoxygenase juxtaglomerular specialized smooth muscle cells release rennin when they are exposed to PGI₂. Glomerular originated synthesis of PGI₂, PGE₂ and TXA₂ are also stimulated by angiotensin II too (Schrier 2013). These three affect glomerular size: While PGI₂ and PGE₂ expands the glomerulus, TXA₂ leads to the contraction. They also affect ultrafiltration coefficient. TXA₂ which is formed in the glomerulus narrows the downstream efferent arteriole, intensifying the resistance and managing glomerular capillary pressure. Interstitial cells of deep renal medulla generate high amounts of PGE₂ when subjected to angiotensin II or vasopressin. Moreover, an increase in renal artery perfusion pressure provokes the increase of PGE₂ production. PGE₂ reduces sodium Henle's reabsorption in loop, hence undermining sodium retention of angiotensin II and subscribing to pressure natriuresis. Ultimately, when collecting duct cells are exposed to vasopressin, they produce PGE₂ and other eicosanoids (Fischer and Weber 1984).

To summarize, products of COX are formed in both renal cortex and medulla and the rate of their generation increases in response to particular stimuli. Once they are released, they affect renal hemodynamics, renin release, sodium and water excretion, erythropoietin production and aids natriuresis and diuresis (Fischer and Weber 1984).

Effects of NSAIDs on renal hemodynamics

There is a relationship between the level of plasma renin activity and the decline in renal blood flow after COX inhibition. As the pretreatment of plasma renin activity increases, the renal blood flow decreases after NSAIDs (Rossat *et al.* 1999).

In addition to the activation of the renin angiotensin system, other risk factors for NSAID-induced reduction of GFR may exist. Renal glomerular diseases such as; renal systemic lupus erythematosus and nephrotoxic serum nephritis are among these factors. Renal insufficiency caused by excision of renal mass is not affected by NSAIDs. Immune-induced glomerulopathy triggers vasodilator eicosanoid production, which manage GFR. It decreases if glomerular COX is inhibited. However, GFR is not reduce by NSAIDs in the absence of active glomerulonephritis because of not increasing vasodilator eicosanoids (Rossat et al. 1999).

Renal artery stenosis increases NSAID nephrotoxicity caused whereas, renin angiotensin system may trigger and support GFR in the stenotic kidney. However, their impacts in the no stenotic kidney is less presumable. **NSAIDs** could reduce contralateral blood flow because of lessen vasodilator eicosanoid formation or raise it by decreased renin release. So that, bilateral renal artery disease or renal artery stenosis in a kidney might be a risk determinant for NSAID-induced renal failure (Rossat *et al.* 1999).

Progressive age is a dependent risk factor for NSAID-induced renal dysfunction. Prevalence of the activation of reninangiotensin system by various factors increases in elderly people. These various factors include; congestive heart failure, diuretic usage, decreased thirst mechanism or atherosclerotic major renal vascular disease (Rossat *et al.* 1999).

To sum up, NSAIDs have little effect on renal blood flow of GFR in normal kidney. However, when the kidney is overexposed to vasoconstrictor stress. eicosanoid production is increased and this plays an important role in maintaining GFR. Similar occurs during active glomerular case inflammation and in the case of renal artery Indomethacin stenosis. can produce profound and prolonged depression of GFR (Loudon et al. 1997).

Effects of NSAID on renin release

There are two clinically convenient consequences of decreased renin release: Reduction of blood pressure and hyperkalemia. In many conditions, it is reported that NSAIDs either do not affect or nebulously increase the blood pressure. Nevertheless, since blood pressure and very high rennin plasma activity are dependent to each other, NSAIDs might reduce blood pressure indeed. The renal baroreceptor mechanism that leads to increased renin synthesis and release when renal perfusion pressure is deducted is enhanced by PGI₂ synthesis. Therefore, blood pressure reduce can be ensured by reducing renin by inhibiting COX. Another common and lifethreatening condition is hyperkalemia (Harris 2003). Prolonged NSAID treatment potassium increases serum levels respectably that peaks at 3-7 days of treatment. Levels of potassium return to normal in 1 or 2 days after withdrawal of NSAIDs. Another risk factor for NSAIDinduced hyperkalemia is diabetes mellitus where; there is a deficiency of insulin mediated cellular potassium reuptake and angiotensin-converting enzyme inhibitor treatment which reduces angiotensin II levels. As a result, NSAIDs continuously decrease basal and stimulated plasma renin activity and plasma aldosterone. In normal functioning kidneys, **NSAIDs** cause remarkable increase in serum potassium. In the case of renal insufficiency or renal insufficiency additional to diabetes mellitus, the extent of hyperkalemia turn out lethal conditions (Stichtenoth et al. 1998).

NSAIDs and sodium and water metabolism

Inhibition of renal PG synthesis can affect sodium excretion by various mechanisms which are shown in Table 1 (Hao *et al.* 2000). PGs have both direct and indirect effects on sodium excretion. Direct impacts are supposedly restricted to the distal nephron, likely to the collecting duct epithelia. Indirect effects, running through hemodynamic and Starling forces. alterations in medullary interstitial pressure or salt content, varied concentrations of other determinants such as; angiotensin II and vasopressin may define the whole outcomes of **NSAIDs** on sodium metabolism. Depending on the potency of NSAID used, sodium excretion would increase, remain unaffected or decrease. In a research, it was reported that; 75 mg single dose of indomethacin lowered urine PGE₂ by 65% and urinary sodium from 200 to 125 mmol/day after 24 hours in high sodium diet and from 43 to 21 mmol/day in low sodium diet. In prolonged studies, indomethacin was reported to cause weight gain by 1 and 2% averagely but rarely up to 5%. Formed fluid retention is in charge of suppressing basal plasma renin activity. In patients with inherent renal disease or disorders where circulating vasoconstrictors are increased, NSAIDs withdrawal lead to fluid retention. Also, reductions in renal blood flow and GFR reduces the filtered load sodium. Water retention generating hyponatremia is occasionally an adverse effect of NSAIDs. However unexpectedly; PGE₂ inhibits the

effect of vasopressin on the collecting duct and NSAIDs may increase the discharge of vasopressin during the volume contraction Furthermore, stimulus. decreasing medullary blood flow may lead to raised osmolality in the medulla and increased reabsorption water which can cause promoted water reabsorption from the collecting duct when administered NSAIDs actually causes serious hyponatremia. A study which was performed among normal subjects showed that 75 mg indomethacin for a week and for 42 days do not cause any change on serum sodium levels whereas, patients who were previously exposed to hyponatremia or those with severe congestive heart failure, cirrhosis with ascites, or who were taking diuretics or with nephrotic syndrome might possibly develop severe hyponatremia by the over usage of NSAIDs (Waddington et al. 2014).

Effect	Mechanism	Effect on sodium excretion
Renal blood flow	Proximal convoluted tubule reabsorption	decrease
Rennin	The ascending loop of Henle reabsorption	increase
Interstitial pressure	The ascending loop of Henle reabsorption	decrease
Medullary blood flow	Addition of Na to tubular fluid	increase
PGE effect on collecting tubules	Collecting duct reabsorption	decrease
Natriuretic effect of AVR	The ascending loop of Henle reabsorption	decrease

Table 1: Possible effects of cyclooxygenase inhibition on sodium excretion.

Effects of NSAIDs on other renal functions

As mentioned before, there is a direct relation between erythropoietin synthesis, PGE₂ synthesis and renal hypoxia. Hypoxia reduces adenosine triphosphate (ATP) which possibly prevent reacylation of arachidonic acid, subsequently allowing it to be more metabolized by COX, leading an increase in the production of PGE₂. PGE₂ increases cAMP by increasing the activity of

adenyl cyclase and activates the phosphorylation of protein kinases (PKs). Transcription of erythropoietin gene can elevate during all stages. It is therefore probable that NSAID-induced deterioration of circulating erythropoietin may lead to anemia usually observed in patients having these drugs (Borda 1992).

DISCUSSION

This present review focused on renal toxic mechanisms induced by NSAIDs. NSAIDs induced nephrotoxic mechanisms including, effects of NSAIDs on renal hemodynamics, renin release and sodium and water metabolism were clarified. The use of NSAID for a long period of time and in large quantities results in the formation of renal papillary necrosis and interstitial nephritis. This pathological condition is called analgesic nephropathy. This review will raise awareness of the patients on the severity of the problem and the importance of reasonable and safe usage of NSAIDs. of NSAIDs Adverse drug reactions

especially on kidney should never be underestimated for protection from serious irreversible risks.

Pharmacists and physicians should sufficiently be informed and be aware of the seriousness of analgesic nephropathy in order to avoid the unnecessary usage of NSAIDs, both with prescriptions and without prescription. The geriatric population is the most vulnerable group to toxic effects of NSAIDs. These people feel more pain and in order to reduce the pain they take NSAIDs in high doses for a long period of time. Moreover. geriatric individuals use many different drugs which can show interactions with NSAIDs. Overall, this population should be well informed and monitored by healthcare professionals in order to decrease the risk of adverse related to NSAIDs.

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